

# N,N'-Bis-(2-hydroxyethyl)-oxamide

<b>Other names:</b>	Ethanediamide, N,N'-bis(2-hydroxyethyl)- N,N'-bis(2-hydroxyethyl)oxamide N,N-Bis(2-hydroxyethyl)oxamide Oxamide, N,N'-bis(2-hydroxyethyl)-
<b>Inchi:</b>	InChI=1S/C6H12N2O4/c9-3-1-7-5(11)6(12)8-2-4-10/h9-10H,1-4H2,(H,7,11)(H,8,12)
<b>InchiKey:</b>	FPQJEXTVQZHURJ-UHFFFAOYSA-N
<b>Formula:</b>	C6H12N2O4
<b>SMILES:</b>	O=C(NCCO)C(=O)NCCO
<b>Mol. weight [g/mol]:</b>	176.17
<b>CAS:</b>	1871-89-2

## Physical Properties

Property code	Value	Unit	Source
gf	-353.06	kJ/mol	Joback Method
hf	-589.85	kJ/mol	Joback Method
hfus	32.87	kJ/mol	Joback Method
hvap	88.67	kJ/mol	Joback Method
log10ws	-0.11		Aqueous Solubility Prediction Method
logp	-2.797		Crippen Method
mcvol	130.240	ml/mol	McGowan Method
pc	4704.19	kPa	Joback Method
tb	729.12	K	Joback Method
tc	909.69	K	Joback Method
tf	484.20	K	Joback Method
vc	0.491	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	365.63	J/molxK	729.12	Joback Method
cpg	373.03	J/molxK	759.21	Joback Method
cpg	380.00	J/molxK	789.31	Joback Method
cpg	386.56	J/molxK	819.40	Joback Method

cpg	392.72	J/mol×K	849.50	Joback Method
cpg	398.50	J/mol×K	879.59	Joback Method
cpg	403.91	J/mol×K	909.69	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1871892&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1871892&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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