

# 2-Methyl-3-methylene-bicyclo[2.2.1]heptane, cis

Inchi:	InChI=1S/C9H14/c1-6-7(2)9-4-3-8(6)5-9/h7-9H,1,3-5H2,2H3/t7-,8?,9?/m0/s1
InchiKey:	ZKASMXOELCEADR-UEJVZZJDSA-N
Formula:	C9H14
SMILES:	C=C1C2CCC(C2)C1C
Mol. weight [g/mol]:	122.21

## Physical Properties

Property code	Value	Unit	Source
gf	179.67	kJ/mol	Joback Method
hf	-25.75	kJ/mol	Joback Method
hfus	13.15	kJ/mol	Joback Method
hvap	35.48	kJ/mol	Joback Method
log10ws	-2.51		Crippen Method
logp	2.609		Crippen Method
mcvol	111.650	ml/mol	McGowan Method
pc	3049.04	kPa	Joback Method
rinpol	1013.00		NIST Webbook
ripol	1085.00		NIST Webbook
tb	417.56	K	Joback Method
tc	618.35	K	Joback Method
tf	232.99	K	Joback Method
vc	0.428	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	228.77	J/molxK	417.56	Joback Method
cpg	245.70	J/molxK	451.03	Joback Method
cpg	261.67	J/molxK	484.49	Joback Method
cpg	276.71	J/molxK	517.96	Joback Method
cpg	290.89	J/molxK	551.42	Joback Method
cpg	304.23	J/molxK	584.89	Joback Method
cpg	316.80	J/molxK	618.35	Joback Method
dvisc	0.0004793	Paxs	232.99	Joback Method

dvisc	0.0004941	Paxs	263.75	Joback Method
dvisc	0.0005061	Paxs	294.51	Joback Method
dvisc	0.0005160	Paxs	325.27	Joback Method
dvisc	0.0005244	Paxs	356.04	Joback Method
dvisc	0.0005315	Paxs	386.80	Joback Method
dvisc	0.0005377	Paxs	417.56	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R246623&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R246623&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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