

# Dimethylmalonic acid, hexadecyl 2-methylpent-3-yl ester

Inchi:	InChI=1S/C27H52O4/c1-7-9-10-11-12-13-14-15-16-17-18-19-20-21-22-30-25(28)27(5,6)
InchiKey:	FGOALVWKTQAWFY-UHFFFAOYSA-N
Formula:	C27H52O4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)C(C)(C)C(=O)OC(CC)C(C)C
Mol. weight [g/mol]:	440.70

## Physical Properties

Property code	Value	Unit	Source
gf	-293.42	kJ/mol	Joback Method
hf	-1109.52	kJ/mol	Joback Method
hfus	56.80	kJ/mol	Joback Method
hvap	91.94	kJ/mol	Joback Method
log10ws	-8.48		Crippen Method
logp	8.015		Crippen Method
mcvol	406.170	ml/mol	McGowan Method
pc	737.22	kPa	Joback Method
rinsol	2756.00		NIST Webbook
tb	965.63	K	Joback Method
tc	1186.37	K	Joback Method
tf	510.79	K	Joback Method
vc	1.573	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1410.02	J/molxK	965.63	Joback Method
cpg	1431.21	J/molxK	1002.42	Joback Method
cpg	1450.77	J/molxK	1039.21	Joback Method
cpg	1468.78	J/molxK	1076.00	Joback Method
cpg	1485.31	J/molxK	1112.79	Joback Method
cpg	1500.44	J/molxK	1149.58	Joback Method
cpg	1514.25	J/molxK	1186.37	Joback Method
dvisc	0.0004279	Paxs	510.79	Joback Method
dvisc	0.0001593	Paxs	586.60	Joback Method

dvisc	0.0000744	Paxs	662.40	Joback Method
dvisc	0.0000406	Paxs	738.21	Joback Method
dvisc	0.0000248	Paxs	814.02	Joback Method
dvisc	0.0000165	Paxs	889.82	Joback Method
dvisc	0.0000117	Paxs	965.63	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U361801&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U361801&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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