

# 5-tert-Butyl-4,6-dinitro-1,2,3-trimethylbenzene

<b>Other names:</b>	Benzene, 1-(1,1-dimethylethyl)-3,4,5-trimethyl-2,6-dinitro- Benzene, 1-(1,1-dimethylethyl)-2,6-dinitro-3,4,5-trimethyl- Benzene, 1-tert-butyl-2,6-dinitro-3,4,5-trimethyl- Benzene, 1-tert-butyl-3,4,5-trimethyl-2,6-dinitro- Musk tibetene Tibetene musk 5-tert-Butyl-1,2,3-trimethyl-4,6-dinitrobenzene 5-t-Butyl-4,6-dinitro-1,2,3-trimethylbenzene Musk tibetine NSC 78470 1-tert-Butyl-2,6-dinitro-3,4,5-trimethylbenzene 1-tert-butyl-3,4,5-trimethyl-2,6-dinitrobenzene
<b>Inchi:</b>	InChI=1S/C13H18N2O4/c1-7-8(2)11(14(16)17)10(13(4,5)6)12(9(7)3)15(18)19/h1-6H3
<b>InchiKey:</b>	MINYPECWDZURGR-UHFFFAOYSA-N
<b>Formula:</b>	C13H18N2O4
<b>SMILES:</b>	Cc1c(C)c([N+](=O)[O-])c(C(C)(C)C)c([N+](=O)[O-])c1C
<b>Mol. weight [g/mol]:</b>	266.29
<b>CAS:</b>	145-39-1

## Physical Properties

Property code	Value	Unit	Source
gf	196.78	kJ/mol	Joback Method
hf	-162.74	kJ/mol	Joback Method
hfus	36.83	kJ/mol	Joback Method
hvap	82.00	kJ/mol	Joback Method
log10ws	-5.51		Crippen Method
logp	3.726		Crippen Method
mcvol	205.110	ml/mol	McGowan Method
pc	2129.52	kPa	Joback Method
rinpol	1907.30		NIST Webbook
tb	848.87	K	Joback Method
tc	1103.46	K	Joback Method
tf	614.93	K	Joback Method
vc	0.808	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	608.06	J/mol×K	848.87	Joback Method
cpg	621.19	J/mol×K	891.30	Joback Method
cpg	633.27	J/mol×K	933.73	Joback Method
cpg	644.37	J/mol×K	976.16	Joback Method
cpg	654.57	J/mol×K	1018.59	Joback Method
cpg	663.94	J/mol×K	1061.03	Joback Method
cpg	672.54	J/mol×K	1103.46	Joback Method

## Sources

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C145391&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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