

# Isobutyl 2-methylvalerate

<b>Other names:</b>	Pentanoic acid, 2-methyl, 2-methylpropyl ester Isobutyl 2-methylpentanoate
<b>Inchi:</b>	InChI=1S/C10H20O2/c1-5-6-9(4)10(11)12-7-8(2)3/h8-9H,5-7H2,1-4H3
<b>InchiKey:</b>	VSNXGQJZBDUEOD-UHFFFAOYSA-N
<b>Formula:</b>	C10H20O2
<b>SMILES:</b>	CCCC(C)C(=O)OCC(C)C
<b>Mol. weight [g/mol]:</b>	172.26
<b>CAS:</b>	6297-42-3

## Physical Properties

Property code	Value	Unit	Source
gf	-205.48	kJ/mol	Joback Method
hf	-505.09	kJ/mol	Joback Method
hfus	17.40	kJ/mol	Joback Method
hvap	46.23	kJ/mol	Joback Method
log10ws	-2.39		Crippen Method
logp	2.622		Crippen Method
mcvol	159.200	ml/mol	McGowan Method
pc	2233.41	kPa	Joback Method
rinpol	1064.00		NIST Webbook
rinpol	1064.00		NIST Webbook
rinpol	1064.00		NIST Webbook
rinpol	1064.00		NIST Webbook
ripol	1417.00		NIST Webbook
ripol	1417.00		NIST Webbook
tb	503.61	K	Joback Method
tc	682.96	K	Joback Method
tf	244.62	K	Joback Method
vc	0.608	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	368.51	J/molxK	503.61	Joback Method

cpg	437.08	J/mol×K	653.07	Joback Method
cpg	424.51	J/mol×K	623.18	Joback Method
cpg	411.38	J/mol×K	593.28	Joback Method
cpg	397.67	J/mol×K	563.39	Joback Method
cpg	383.39	J/mol×K	533.50	Joback Method
cpg	449.08	J/mol×K	682.96	Joback Method
dvisc	0.0001994	Paxs	503.61	Joback Method
dvisc	0.0002734	Paxs	460.44	Joback Method
dvisc	0.0004002	Paxs	417.28	Joback Method
dvisc	0.0006396	Paxs	374.12	Joback Method
dvisc	0.0011553	Paxs	330.95	Joback Method
dvisc	0.0024917	Paxs	287.79	Joback Method
dvisc	0.0070487	Paxs	244.62	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6297423&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6297423&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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