

1,2-Cyclohexanediol, trans-

Other names:	(1R,2R)-1,2-cyclohexanediol 1,2-Cyclohexanediol, (E)- 1,2-trans-Cyclohexanediol trans-1,2-Cyclohexanediol trans-1,2-Dihydroxycyclohexane trans-cyclohexane-1,2-diol
Inchi:	InChI=1S/C6H12O2/c7-5-3-1-2-4-6(5)8/h5-8H,1-4H2/t5-,6-/m1/s1
InchiKey:	PFURGBBHAXLIO-PHDIDXHHSA-N
Formula:	C6H12O2
SMILES:	OC1CCCCC1O
Mol. weight [g/mol]:	116.16
CAS:	1460-57-7

Physical Properties

Property code	Value	Unit	Source
chs	-3520.00 ± 0.80	kJ/mol	NIST Webbook
gf	-257.26	kJ/mol	Joback Method
hf	-437.65	kJ/mol	Joback Method
hfus	12.38	kJ/mol	Joback Method
hvap	62.43	kJ/mol	Joback Method
log10ws	-0.98		Crippen Method
logp	0.282		Crippen Method
mcvol	96.280	ml/mol	McGowan Method
pc	4835.95	kPa	Joback Method
tb	535.92	K	Joback Method
tc	720.36	K	Joback Method
tf	376.40	K	Solid-Liquid Equilibria of the trans-1,2-Cyclohexanediol + Ethyl Acetate + Water Ternary System
tf	376.65 ± 1.50	K	NIST Webbook
tf	381.00	K	A Novel Co-crystal of (R,R)-1,2-Cyclohexanediol with (R,R)-Tartaric Acid, a Key Structure for Resolution of the Racemic (+-)-trans-Diol in Supercritical Carbon Dioxide, and the Binary Melting Phase Relations.

vc	0.342	m3/kmol	Joback Method
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Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	241.23	J/mol×K	535.92	Joback Method
cpg	252.43	J/mol×K	566.66	Joback Method
cpg	263.09	J/mol×K	597.40	Joback Method
cpg	273.22	J/mol×K	628.14	Joback Method
cpg	282.84	J/mol×K	658.88	Joback Method
cpg	291.95	J/mol×K	689.62	Joback Method
cpg	300.56	J/mol×K	720.36	Joback Method
dvisc	0.0000802	Pa×s	535.92	Joback Method
dvisc	0.0131225	Pa×s	324.45	Joback Method
dvisc	0.0029585	Pa×s	366.75	Joback Method
dvisc	0.0909733	Pa×s	282.16	Joback Method
dvisc	0.0003475	Pa×s	451.33	Joback Method
dvisc	0.0001568	Pa×s	493.63	Joback Method
dvisc	0.0009076	Pa×s	409.04	Joback Method
hfust	21.00	kJ/mol	382.60	NIST Webbook
hfust	16.37	kJ/mol	375.70	NIST Webbook
hfust	18.51	kJ/mol	372.30	NIST Webbook
hfust	18.51	kJ/mol	372.30	NIST Webbook
hsubt	85.90 ± 1.40	kJ/mol	343.00	NIST Webbook
hsubt	42.50	kJ/mol	304.50	NIST Webbook

Sources

Crippen Method:
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:
https://www.chemeo.com/doc/models/crippen_log10ws

A Novel Co-crystal of (R,R)-1,2-Cyclohexanediol with Propylene Glycol: Crystal Structure, Solid-Liquid Equilibrium, and Thermodynamic Properties
<https://www.doi.org/10.1016/j.tca.2009.09.001>

Solid-Liquid Equilibrium of the System 1,2-Cyclohexanediol + Ethyl Acrylate
<https://www.doi.org/10.1021/je050338e>

Joback Method
https://en.wikipedia.org/wiki/Joback_method

A Simple and Accurate Method for Estimating the Heat of Vaporization of Organic Compounds
<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:
<http://webbook.nist.gov/cgi/cbook.cgi?ID=C1460577&Units=SI>

Legend

chs:	Standard solid enthalpy of combustion
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hsubt:	Enthalpy of sublimation at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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