

# 2-[2-[2-(2-Butoxyethoxy)ethoxy]ethoxy]ethyl 2,2,3,3,3-pentafluoropropanoate

**Other names:** Tetraethylene glycol monobutyl ether, pentafluoropropionate

3,6,9,12-Tetraoxahexadec-1-yl pentafluoropropionate

**Inchi:** InChI=1S/C15H25F5O6/c1-2-3-4-22-5-6-23-7-8-24-9-10-25-11-12-26-13(21)14(16,17)15

**InchiKey:** COFNBS PQYZUH QI-UHFFFAOYSA-N

**Formula:** C15H25F5O6

**SMILES:** CCCCOC COCCOCCOCCOC(=O)C(F)(F)C(F)(F)F

**Mol. weight [g/mol]:** 396.35

## Physical Properties

Property code	Value	Unit	Source
gf	-1546.87	kJ/mol	Joback Method
hf	-2124.66	kJ/mol	Joback Method
hfus	42.72	kJ/mol	Joback Method
h vap	61.10	kJ/mol	Joback Method
log10ws	-2.29		Crippen Method
logp	2.594		Crippen Method
m cvol	261.980	ml/mol	McGowan Method
pc	1216.59	kPa	Joback Method
rinpol	1703.30		NIST Webbook
tb	698.46	K	Joback Method
tc	860.21	K	Joback Method
tf	427.68	K	Joback Method
vc	1.040	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	785.15	J/mol×K	698.46	Joback Method
cpg	800.34	J/mol×K	725.42	Joback Method
cpg	814.77	J/mol×K	752.38	Joback Method
cpg	828.45	J/mol×K	779.34	Joback Method
cpg	841.39	J/mol×K	806.30	Joback Method
cpg	853.59	J/mol×K	833.25	Joback Method
cpg	865.05	J/mol×K	860.21	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U351989&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U351989&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/42-708-9/2-2-2-2-2-Butoxyethoxy-ethoxy-ethyl-2-2-3-3-3-pentafluoropropanoate.p>

Generated by Cheméo on 2024-04-24 18:25:25.795425802 +0000 UTC m=+16272374.716003124.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.