

# Diethylmalonic acid, 4-chlorobenzyl ethyl ester

Inchi:	InChI=1S/C16H21ClO4/c1-4-16(5-2,14(18)20-6-3)15(19)21-11-12-7-9-13(17)10-8-12/h7-
InchiKey:	GBYJIXOOZLRTES-UHFFFAOYSA-N
Formula:	C16H21ClO4
SMILES:	CCOC(=O)C(CC)(CC)C(=O)OCc1ccc(Cl)cc1
Mol. weight [g/mol]:	312.79

## Physical Properties

Property code	Value	Unit	Source
gf	-290.31	kJ/mol	Joback Method
hf	-662.60	kJ/mol	Joback Method
hfus	33.20	kJ/mol	Joback Method
hvap	75.55	kJ/mol	Joback Method
log10ws	-4.29		Crippen Method
logp	3.753		Crippen Method
mcvol	239.660	ml/mol	McGowan Method
pc	1789.39	kPa	Joback Method
rinsol	1997.00		NIST Webbook
tb	783.92	K	Joback Method
tc	997.75	K	Joback Method
tf	485.68	K	Joback Method
vc	0.909	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	677.07	J/molxK	783.92	Joback Method
cpg	738.11	J/molxK	962.11	Joback Method
cpg	727.87	J/molxK	926.47	Joback Method
cpg	716.69	J/molxK	890.83	Joback Method
cpg	704.51	J/molxK	855.20	Joback Method
cpg	691.32	J/molxK	819.56	Joback Method
cpg	747.43	J/molxK	997.75	Joback Method
dvisc	0.0000669	Paxs	783.92	Joback Method
dvisc	0.0000863	Paxs	734.21	Joback Method

dvisc	0.0001155	Paxs	684.51	Joback Method
dvisc	0.0001618	Paxs	634.80	Joback Method
dvisc	0.0002400	Paxs	585.09	Joback Method
dvisc	0.0003832	Paxs	535.39	Joback Method
dvisc	0.0006732	Paxs	485.68	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369774&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369774&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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