

# Amorpha-4,9-dien-2-ol

<b>Inchi:</b>	InChI=1S/C15H24O/c1-9(2)12-6-5-11(4)15-13(12)7-10(3)8-14(15)16/h5,7,9,12-16H,6,8H
<b>InchiKey:</b>	STLBTFMCSOXEAQ-NAOUJUTFSA-N
<b>Formula:</b>	C15H24O
<b>SMILES:</b>	CC1=CC2C(C(C)C)CC=C(C)C2C(O)C1
<b>Mol. weight [g/mol]:</b>	220.35

## Physical Properties

Property code	Value	Unit	Source
gf	34.50	kJ/mol	Joback Method
hf	-337.54	kJ/mol	Joback Method
hfus	26.85	kJ/mol	Joback Method
hvap	67.08	kJ/mol	Joback Method
log10ws	-4.01		Crippen Method
logp	3.552		Crippen Method
mcvol	197.760	ml/mol	McGowan Method
pc	2025.41	kPa	Joback Method
rinpol	1630.00		NIST Webbook
rinpol	1687.00		NIST Webbook
tb	663.84	K	Joback Method
tc	864.81	K	Joback Method
tf	344.51	K	Joback Method
vc	0.741	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	578.38	J/molxK	663.84	Joback Method
cpg	662.51	J/molxK	831.32	Joback Method
cpg	647.70	J/molxK	797.82	Joback Method
cpg	631.91	J/molxK	764.33	Joback Method
cpg	615.12	J/molxK	730.83	Joback Method
cpg	597.29	J/molxK	697.34	Joback Method
cpg	676.39	J/molxK	864.81	Joback Method
dvisc	0.0001311	Paxs	663.84	Joback Method

dvisc	0.0001823	Paxs	610.62	Joback Method
dvisc	0.0002700	Paxs	557.40	Joback Method
dvisc	0.0004346	Paxs	504.17	Joback Method
dvisc	0.0007826	Paxs	450.95	Joback Method
dvisc	0.0016494	Paxs	397.73	Joback Method
dvisc	0.0043769	Paxs	344.51	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R281626&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R281626&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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