

# 2-Ethylhexyl tridecanoate

<b>Inchi:</b>	InChI=1S/C21H42O2/c1-4-7-9-10-11-12-13-14-15-16-18-21(22)23-19-20(6-3)17-8-5-2/h
<b>InchiKey:</b>	VLUCLKQLSQKOPQ-UHFFFAOYSA-N
<b>Formula:</b>	C21H42O2
<b>SMILES:</b>	CCCCCCCCCCCC(=O)OCC(CC)CCCC
<b>Mol. weight [g/mol]:</b>	326.56

## Physical Properties

Property code	Value	Unit	Source
gf	-110.42	kJ/mol	Joback Method
hf	-726.85	kJ/mol	Joback Method
hfus	49.41	kJ/mol	Joback Method
hvap	71.11	kJ/mol	Joback Method
log10ws	-7.23		Crippen Method
logp	7.057		Crippen Method
mcvol	314.190	ml/mol	McGowan Method
pc	988.88	kPa	Joback Method
rinpol	2181.00		NIST Webbook
tb	755.73	K	Joback Method
tc	930.28	K	Joback Method
tf	383.59	K	Joback Method
vc	1.230	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	969.75	J/molxK	755.73	Joback Method
cpg	1061.34	J/molxK	901.19	Joback Method
cpg	1044.89	J/molxK	872.09	Joback Method
cpg	1027.53	J/molxK	843.00	Joback Method
cpg	1009.24	J/molxK	813.91	Joback Method
cpg	989.99	J/molxK	784.82	Joback Method
cpg	1076.91	J/molxK	930.28	Joback Method
dvisc	0.0000590	Paxs	755.73	Joback Method
dvisc	0.0000811	Paxs	693.71	Joback Method

dvisc	0.0001187	Paxs	631.68	Joback Method
dvisc	0.0001889	Paxs	569.66	Joback Method
dvisc	0.0003366	Paxs	507.64	Joback Method
dvisc	0.0007045	Paxs	445.61	Joback Method
dvisc	0.0018724	Paxs	383.59	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R540435&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R540435&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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