

# Cyclohexyl propionate

<b>Other names:</b>	Cyclohexyl n-propionate Propionic acid cyclohexyl ester cyclohexyl propanoate propanoic acid, cyclohexyl ester propionic acid, cyclohexyl ester
<b>Inchi:</b>	InChI=1S/C9H16O2/c1-2-9(10)11-8-6-4-3-5-7-8/h8H,2-7H2,1H3
<b>InchiKey:</b>	MAMMVUWCKMOLSG-UHFFFAOYSA-N
<b>Formula:</b>	C9H16O2
<b>SMILES:</b>	CCC(=O)OC1CCCCC1
<b>Mol. weight [g/mol]:</b>	156.22
<b>CAS:</b>	6222-35-1

## Physical Properties

Property code	Value	Unit	Source
gf	-184.57	kJ/mol	Joback Method
hf	-419.57	kJ/mol	Joback Method
hfus	13.69	kJ/mol	Joback Method
hvap	56.80	kJ/mol	NIST Webbook
hvap	55.90 ± 0.10	kJ/mol	NIST Webbook
hvap	59.40 ± 0.80	kJ/mol	NIST Webbook
hvap	56.40 ± 0.50	kJ/mol	NIST Webbook
hvap	54.30 ± 0.40	kJ/mol	NIST Webbook
log10ws	-2.46		Crippen Method
logp	2.272		Crippen Method
mvol	134.250	ml/mol	McGowan Method
pc	2982.79	kPa	Joback Method
rinpol	1120.00		NIST Webbook
rinpol	1111.00		NIST Webbook
ripol	1408.00		NIST Webbook
tb	501.16	K	Joback Method
tc	709.61	K	Joback Method
tf	270.73	K	Joback Method
vc	0.496	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	398.17	J/molxK	709.61	Joback Method
cpg	308.34	J/molxK	501.16	Joback Method
cpg	325.39	J/molxK	535.90	Joback Method
cpg	341.59	J/molxK	570.64	Joback Method
cpg	356.96	J/molxK	605.38	Joback Method
cpg	371.51	J/molxK	640.12	Joback Method
cpg	385.24	J/molxK	674.87	Joback Method
dvisc	0.0002611	Paxs	501.16	Joback Method
dvisc	0.0043071	Paxs	270.73	Joback Method
dvisc	0.0020196	Paxs	309.13	Joback Method
dvisc	0.0011196	Paxs	347.54	Joback Method
dvisc	0.0006979	Paxs	385.94	Joback Method
dvisc	0.0004740	Paxs	424.35	Joback Method
dvisc	0.0003432	Paxs	462.75	Joback Method
hvapt	55.73	kJ/mol	300.55	Comprehensive Study of Vapor Pressures and Enthalpies of Vaporization of Cyclohexyl Esters

## Sources

Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Comprehensive Study of Vapor Pressures and Enthalpies of Vaporization of Cyclohexyl Esters:	<a href="https://www.doi.org/10.1021/je025634v">https://www.doi.org/10.1021/je025634v</a>
Vaporization Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6222351&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6222351&amp;Units=SI</a>

## Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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