

1-Bromo-1-chloroethane

Other names:	CH3CHClBr Ethane, 1-bromo-1-chloro-
Inchi:	InChI=1S/C2H4BrCl/c1-2(3)4/h2H,1H3
InchiKey:	QMSVNSDEZTYAS-UHFFFAOYSA-N
Formula:	C2H4BrCl
SMILES:	CC(Cl)Br
Mol. weight [g/mol]:	143.41
CAS:	593-96-4

Physical Properties

Property code	Value	Unit	Source
gf	-34.09	kJ/mol	Joback Method
hf	-81.00 ± 3.00	kJ/mol	NIST Webbook
hfus	6.90	kJ/mol	Joback Method
hvap	30.48	kJ/mol	Joback Method
ie	10.37	eV	NIST Webbook
ie	10.42	eV	NIST Webbook
log10ws	-1.85		Crippen Method
logp	1.966		Crippen Method
mcvol	68.780	ml/mol	McGowan Method
pc	5327.93	kPa	Joback Method
tb	356.20	K	NIST Webbook
tc	549.86	K	Joback Method
tf	187.02	K	Joback Method
vc	0.253	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	86.20	J/molxK	348.31	Joback Method
cpg	90.75	J/molxK	381.90	Joback Method
cpg	95.03	J/molxK	415.49	Joback Method
cpg	99.07	J/molxK	449.09	Joback Method
cpg	102.86	J/molxK	482.68	Joback Method

cpg	106.43	J/molxK	516.27	Joback Method
cpg	109.79	J/molxK	549.86	Joback Method
dvisc	0.0025302	Paxs	213.90	Joback Method
dvisc	0.0049301	Paxs	187.02	Joback Method
dvisc	0.0015071	Paxs	240.78	Joback Method
dvisc	0.0009961	Paxs	267.66	Joback Method
dvisc	0.0007101	Paxs	294.55	Joback Method
dvisc	0.0005357	Paxs	321.43	Joback Method
dvisc	0.0004221	Paxs	348.31	Joback Method
hvapt	33.10	kJ/mol	323.00	NIST Webbook
hvapt	46.70	kJ/mol	296.50	NIST Webbook

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	354.10 ± 0.50	K	98.00	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.18014e+01
Coeff. B	-1.77343e+03
Coeff. C	-1.09259e+02
Temperature range (K), min.	263.29
Temperature range (K), max.	382.52

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C593964&Units=SI

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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