

# Diethylmalonic acid, 4-biphenyl undecyl ester

<b>Inchi:</b>	InChI=1S/C30H42O4/c1-4-7-8-9-10-11-12-13-17-24-33-28(31)30(5-2,6-3)29(32)34-27-22
<b>InchiKey:</b>	UJLOIQCTYHXFCN-UHFFFAOYSA-N
<b>Formula:</b>	C30H42O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccc(-c2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	466.65

## Physical Properties

Property code	Value	Unit	Source
gf	-48.09	kJ/mol	Joback Method
hf	-699.29	kJ/mol	Joback Method
hfus	59.31	kJ/mol	Joback Method
hvap	104.60	kJ/mol	Joback Method
log10ws	-9.71		Crippen Method
logp	8.139		Crippen Method
mcvol	400.920	ml/mol	McGowan Method
pc	896.41	kPa	Joback Method
rinpol	3395.00		NIST Webbook
rinpol	3395.00		NIST Webbook
tb	1093.49	K	Joback Method
tc	1339.37	K	Joback Method
tf	639.96	K	Joback Method
vc	1.536	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1385.18	J/molxK	1093.49	Joback Method
cpg	1449.45	J/molxK	1298.39	Joback Method
cpg	1439.03	J/molxK	1257.41	Joback Method
cpg	1427.52	J/molxK	1216.43	Joback Method
cpg	1414.80	J/molxK	1175.45	Joback Method
cpg	1400.73	J/molxK	1134.47	Joback Method
cpg	1458.92	J/molxK	1339.37	Joback Method
dvisc	0.0000094	Paxs	1093.49	Joback Method

dvisc	0.0000125	Paxs	1017.90	Joback Method
dvisc	0.0000172	Paxs	942.31	Joback Method
dvisc	0.0000253	Paxs	866.73	Joback Method
dvisc	0.0000398	Paxs	791.14	Joback Method
dvisc	0.0000691	Paxs	715.55	Joback Method
dvisc	0.0001366	Paxs	639.96	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370431&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370431&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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