

exo-5-Ethenyl-5-methyl-2-norbornene

Inchi: InChI=1S/C10H14/c1-3-10(2)7-8-4-5-9(10)6-8/h3-5,8-9H,1,6-7H2,2H3/t8-,9+,10+/m1/s1
InchiKey: SVLPZOWDVPMLAH-UTLUCORTSA-N
Formula: C10H14
SMILES: C=CC1(C)CC2C=CC1C2
Mol. weight [g/mol]: 134.22

Physical Properties

Property code	Value	Unit	Source
gf	247.32	kJ/mol	Joback Method
hf	67.82	kJ/mol	Joback Method
hfus	10.54	kJ/mol	Joback Method
hvap	36.01	kJ/mol	Joback Method
log10ws	-2.78		Crippen Method
logp	2.775		Crippen Method
mcvol	121.440	ml/mol	McGowan Method
pc	3076.16	kPa	Joback Method
rinpol	923.00		NIST Webbook
tb	437.36	K	Joback Method
tc	649.85	K	Joback Method
tf	253.48	K	Joback Method
vc	0.466	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	256.14	J/mol×K	437.36	Joback Method
cpg	274.30	J/mol×K	472.78	Joback Method
cpg	290.96	J/mol×K	508.19	Joback Method
cpg	306.28	J/mol×K	543.61	Joback Method
cpg	320.41	J/mol×K	579.02	Joback Method
cpg	333.50	J/mol×K	614.44	Joback Method
cpg	345.70	J/mol×K	649.85	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R128080&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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