

1-Azacyclononan-2-one

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|-----------------------------|---|
| Other names: | 1H-Azonin-2(3H)-one, hexahydro- 2-Perhydroazoninone 2-azacyclononanone 2H-Azonin-2-one, octahydro- 8-Aminooctanoic acid lactam 8-Octanelactam Azacyclononan-2-one Capryllactam Caprylolactam NSC 59017 Octahydro-2H-azonin-2-one Octamethylenimine, 2-oxo- Octanoic acid, 8-amino-, lactam cyclooctanone isooxime «eta»-Capryllactam «omega»-Caprylolactam «omega»-Octalactam |
| Inchi: | InChI=1S/C8H15NO/c10-8-6-4-2-1-3-5-7-9-8/h1-7H2,(H,9,10) |
| InchiKey: | YDLSUFFXJYEVHW-UHFFFAOYSA-N |
| Formula: | C8H15NO |
| SMILES: | O=C1CCCCCCCN1 |
| Mol. weight [g/mol]: | 141.21 |
| CAS: | 935-30-8 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|--------|----------------|
| gf | -22.54 | kJ/mol | Joback Method |
| hf | -252.16 | kJ/mol | Joback Method |
| hfus | 10.04 | kJ/mol | Joback Method |
| hvap | 45.66 | kJ/mol | Joback Method |
| ie | 9.12 | eV | NIST Webbook |
| log10ws | -2.03 | | Crippen Method |
| logp | 1.457 | | Crippen Method |
| mcpvol | 124.270 | ml/mol | McGowan Method |
| pc | 3843.54 | kPa | Joback Method |
| tb | 535.84 | K | Joback Method |
| tc | 789.05 | K | Joback Method |

| | | | |
|----|--------|----------------------|---------------|
| tf | 354.23 | K | Joback Method |
| vc | 0.438 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 290.80 | J/mol×K | 535.84 | Joback Method |
| cpg | 311.06 | J/mol×K | 578.04 | Joback Method |
| cpg | 330.25 | J/mol×K | 620.24 | Joback Method |
| cpg | 348.31 | J/mol×K | 662.45 | Joback Method |
| cpg | 365.20 | J/mol×K | 704.65 | Joback Method |
| cpg | 380.86 | J/mol×K | 746.85 | Joback Method |
| cpg | 395.24 | J/mol×K | 789.05 | Joback Method |

Sources

| | |
|--|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C935308&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Enthalpies of reaction of lanthanide trifluoroacetate trihydrate with 2-azacyclononanone to form lanthanide trifluoroacetate-tris 2-azacyclononanone: | https://www.doi.org/10.1016/j.tca.2006.12.017 |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| ie: | Ionization energy |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |

tb: Normal Boiling Point Temperature
tc: Critical Temperature
tf: Normal melting (fusion) point
vc: Critical Volume

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