

# 2-Chlorostyrene

<b>Other names:</b>	1-Chloro-2-vinylbenzene Benzene, 1-chloro-2-ethenyl- Styrene, o-chloro- o-Chlorostyrene
<b>Inchi:</b>	InChI=1S/C8H7Cl/c1-2-7-5-3-4-6-8(7)9/h2-6H,1H2
<b>InchiKey:</b>	ISRGONDNXBCDBM-UHFFFAOYSA-N
<b>Formula:</b>	C8H7Cl
<b>SMILES:</b>	C=Cc1ccccc1Cl
<b>Mol. weight [g/mol]:</b>	138.59
<b>CAS:</b>	2039-87-4

## Physical Properties

Property code	Value	Unit	Source
gf	195.17	kJ/mol	Joback Method
hf	126.30	kJ/mol	Joback Method
hfus	13.04	kJ/mol	Joback Method
hvap	40.06	kJ/mol	Joback Method
log10ws	-2.98		Crippen Method
logp	2.983		Crippen Method
mcvol	107.760	ml/mol	McGowan Method
pc	3607.21	kPa	Joback Method
tb	461.90	K	NIST Webbook
tc	671.47	K	Joback Method
tf	247.02	K	Joback Method
vc	0.406	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	241.74	J/molxK	671.47	Joback Method
cpg	186.15	J/molxK	448.21	Joback Method
cpg	197.08	J/molxK	485.42	Joback Method
cpg	207.30	J/molxK	522.63	Joback Method
cpg	216.84	J/molxK	559.84	Joback Method

cpg	225.74	J/molxK	597.05	Joback Method
cpg	234.03	J/molxK	634.26	Joback Method
dvisc	0.0002538	Paxs	448.21	Joback Method
dvisc	0.0020587	Paxs	247.02	Joback Method
dvisc	0.0011790	Paxs	280.55	Joback Method
dvisc	0.0007605	Paxs	314.08	Joback Method
dvisc	0.0005339	Paxs	347.62	Joback Method
dvisc	0.0003989	Paxs	381.15	Joback Method
dvisc	0.0003124	Paxs	414.68	Joback Method
hvapt	46.00	kJ/mol	443.00	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	332.20	K	0.90	NIST Webbook
tbrp	337.80	K	1.30	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.46526e+01
Coeff. B	-4.16255e+03
Coeff. C	-4.70680e+01
Temperature range (K), min.	336.84
Temperature range (K), max.	492.68

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2039874&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2039874&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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