

# 1,2-Cyclohexanedicarboxylic acid, dimethyl ester, cis-

<b>Other names:</b>	1,2-Cyclohexanedicarboxylic acid, 1,2-dimethyl ester, (1S,2R)-rel-dimethyl cis-1,2-cyclohexanedicarboxylate
<b>Inchi:</b>	InChI=1S/C10H16O4/c1-13-9(11)7-5-3-4-6-8(7)10(12)14-2/h7-8H,3-6H2,1-2H3/t7-,8+
<b>InchiKey:</b>	AIACXWOETVLBIA-OCAPTIKFSA-N
<b>Formula:</b>	C10H16O4
<b>SMILES:</b>	COC(=O)C1CCCCC1C(=O)OC
<b>Mol. weight [g/mol]:</b>	200.23
<b>CAS:</b>	1687-29-2

## Physical Properties

Property code	Value	Unit	Source
gf	-417.78	kJ/mol	Joback Method
hf	-705.35	kJ/mol	Joback Method
hfus	20.14	kJ/mol	Joback Method
hvap	56.29	kJ/mol	Joback Method
log10ws	-1.14		Crippen Method
logp	1.139		Crippen Method
mcvol	155.780	ml/mol	McGowan Method
pc	2729.71	kPa	Joback Method
rinpol	1384.00		NIST Webbook
rinpol	1359.00		NIST Webbook
rinpol	1366.00		NIST Webbook
rinpol	1375.00		NIST Webbook
tb	595.66	K	Joback Method
tc	806.87	K	Joback Method
tf	349.92	K	Joback Method
vc	0.576	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	406.56	J/mol×K	595.66	Joback Method
cpg	480.69	J/mol×K	771.67	Joback Method
cpg	467.69	J/mol×K	736.46	Joback Method

cpg	453.76	J/molxK	701.26	Joback Method
cpg	438.92	J/molxK	666.06	Joback Method
cpg	423.19	J/molxK	630.86	Joback Method
cpg	492.77	J/molxK	806.87	Joback Method
dvisc	0.0002314	Paxs	595.66	Joback Method
dvisc	0.0002916	Paxs	554.70	Joback Method
dvisc	0.0003813	Paxs	513.75	Joback Method
dvisc	0.0005222	Paxs	472.79	Joback Method
dvisc	0.0007592	Paxs	431.83	Joback Method
dvisc	0.0011937	Paxs	390.88	Joback Method
dvisc	0.0020866	Paxs	349.92	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1687292&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1687292&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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