

# 1H-Indole-2,3-dione, 5-methyl-

<b>Other names:</b>	Indole-2,3-dione, 5-methyl- 5-Methylisatin 2,3-Dihydro-5-methylindole-2,3-dione 5-Methylindole-2,3-dione 5-Methyl-2,3-indolinedione 5-methylindole-2,3(1H)-dione
<b>Inchi:</b>	InChI=1S/C9H7NO2/c1-5-2-3-7-6(4-5)8(11)9(12)10-7/h2-4H,1H3,(H,10,11,12)
<b>InchiKey:</b>	VAJCSPZKMQIAP-UHFFFAOYSA-N
<b>Formula:</b>	C9H7NO2
<b>SMILES:</b>	<chem>Cc1ccc2c(c1)C(=O)C(=O)N2</chem>
<b>Mol. weight [g/mol]:</b>	161.16
<b>CAS:</b>	608-05-9

## Physical Properties

Property code	Value	Unit	Source
gf	29.04	kJ/mol	Joback Method
hf	-159.95	kJ/mol	Joback Method
hfus	18.00	kJ/mol	Joback Method
hvap	54.70	kJ/mol	Joback Method
log10ws	-1.87		Crippen Method
logp	1.130		Crippen Method
mcvol	116.170	ml/mol	McGowan Method
pc	4311.22	kPa	Joback Method
tb	637.56	K	Joback Method
tc	901.55	K	Joback Method
tf	506.30	K	Joback Method
vc	0.441	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	282.28	J/mol×K	637.56	Joback Method
cpg	295.12	J/mol×K	681.56	Joback Method
cpg	307.13	J/mol×K	725.56	Joback Method

cpg	318.28	J/mol×K	769.56	Joback Method
cpg	328.56	J/mol×K	813.56	Joback Method
cpg	337.92	J/mol×K	857.55	Joback Method
cpg	346.34	J/mol×K	901.55	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C608059&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C608059&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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