

2-Propanone, 1-(5-methyl-3H-1,2-dithiol-3-ylidene)-

Other names:	3,5-Heptadien-2-one, 4,6-epidithio-
Inchi:	InChI=1S/C7H8OS2/c1-5(8)3-7-4-6(2)9-10-7/h3-4H,1-2H3/b7-3-
InchiKey:	YAIGSGPVNPIUNY-CLTKARDFSA-N
Formula:	C7H8OS2
SMILES:	CC(=O)C=C1C=C(C)SS1
Mol. weight [g/mol]:	172.27
CAS:	1005-55-6

Physical Properties

Property code	Value	Unit	Source
gf	68.91	kJ/mol	Joback Method
hf	-6.71	kJ/mol	Joback Method
hfus	16.82	kJ/mol	Joback Method
hvap	51.85	kJ/mol	Joback Method
ie	7.68	eV	NIST Webbook
log10ws	-3.38		Crippen Method
logp	2.758		Crippen Method
mcvol	124.300	ml/mol	McGowan Method
pc	4026.13	kPa	Joback Method
tb	539.82	K	Joback Method
tc	787.85	K	Joback Method
tf	424.26	K	Joback Method
vc	0.436	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	246.89	J/molxK	539.82	Joback Method
cpg	257.42	J/molxK	581.16	Joback Method
cpg	267.18	J/molxK	622.50	Joback Method
cpg	276.24	J/molxK	663.83	Joback Method
cpg	284.65	J/molxK	705.17	Joback Method
cpg	292.46	J/molxK	746.51	Joback Method
cpg	299.72	J/molxK	787.85	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C1005556&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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