

# Cyclopentanecarboxylic acid, 1-(2-butenyl)-2-oxo-, ethyl ester, (Z)-

Other names:

Ethyl 1-[(2Z)-2-butenyl]-2-oxocyclopentanecarboxylate

cis-Cyclopentanecarboxylic acid, 1-(2-butenyl)-2-oxo-, ethyl ester

Inchi:

InChI=1S/C12H18O3/c1-3-5-8-12(11(14)15-4-2)9-6-7-10(12)13/h3,5H,4,6-9H2,1-2H3/b5

InchiKey:

MUPFBGVUYDDKGG-HYXAFXHYSA-N

Formula:

C12H18O3

SMILES:

CC=CCC1(C(=O)OCC)CCCC1=O

Mol. weight [g/mol]:

210.27

CAS:

131534-41-3

## Physical Properties

Property code	Value	Unit	Source
gf	-195.07	kJ/mol	Joback Method
hf	-480.57	kJ/mol	Joback Method
hfus	16.97	kJ/mol	Joback Method
hvap	54.77	kJ/mol	Joback Method
log10ws	-2.50		Crippen Method
logp	2.255		Crippen Method
mcvol	173.790	ml/mol	McGowan Method
pc	2470.27	kPa	Joback Method
tb	637.75	K	Joback Method
tc	859.37	K	Joback Method
tf	395.10	K	Joback Method
vc	0.657	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	465.25	J/mol×K	637.75	Joback Method
cpg	482.08	J/mol×K	674.69	Joback Method
cpg	498.06	J/mol×K	711.62	Joback Method
cpg	513.29	J/mol×K	748.56	Joback Method
cpg	527.87	J/mol×K	785.50	Joback Method
cpg	541.92	J/mol×K	822.43	Joback Method
cpg	555.52	J/mol×K	859.37	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C131534413&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C131534413&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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