

# di-ethoxylated 1,6 hexane diol diacrylate

Inchi:	InChI=1S/C16H26O6/c1-3-15(17)21-13-11-19-9-7-5-6-8-10-20-12-14-22-16(18)4-2/h3-4H
InchiKey:	GIMNDOUZTMKUHI-UHFFFAOYSA-N
Formula:	C16H26O6
SMILES:	C=CC(=O)OCCOCCCCCOCCOC(=O)C=C
Mol. weight [g/mol]:	314.37

## Physical Properties

Property code	Value	Unit	Source
gf	-418.32	kJ/mol	Joback Method
hf	-876.75	kJ/mol	Joback Method
hfus	42.59	kJ/mol	Joback Method
hvap	73.00	kJ/mol	Joback Method
log10ws	-2.13		Crippen Method
logp	2.038		Crippen Method
mcvol	254.320	ml/mol	McGowan Method
pc	1465.73	kPa	Joback Method
rinsol	2131.00		NIST Webbook
tb	756.26	K	Joback Method
tc	937.80	K	Joback Method
tf	455.34	K	Joback Method
vc	0.978	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	742.55	J/molxK	756.26	Joback Method
cpg	757.66	J/molxK	786.52	Joback Method
cpg	771.92	J/molxK	816.77	Joback Method
cpg	785.32	J/molxK	847.03	Joback Method
cpg	797.85	J/molxK	877.28	Joback Method
cpg	809.53	J/molxK	907.54	Joback Method
cpg	820.33	J/molxK	937.80	Joback Method
dvisc	0.0006178	Paxs	455.34	Joback Method
dvisc	0.0003455	Paxs	505.49	Joback Method

dvisc	0.0002146	Paxs	555.65	Joback Method
dvisc	0.0001442	Paxs	605.80	Joback Method
dvisc	0.0001030	Paxs	655.95	Joback Method
dvisc	0.0000771	Paxs	706.11	Joback Method
dvisc	0.0000601	Paxs	756.26	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R508357&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R508357&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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