

# Benzo[qr]naphtho[2,1,8,7-fghi]pentacene

|                      |                                                                                   |
|----------------------|-----------------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C30H16/c1-2-7-19-18(6-1)16-26-23-11-5-10-22-20-8-3-4-9-21(20)25-15-13-17 |
| InchiKey:            | AVVLVJLIKRMPPQZ-UHFFFAOYSA-N                                                      |
| Formula:             | C30H16                                                                            |
| SMILES:              | c1ccc2c(c1)cc1c3cccc4c5ccccc5c5ccc6ccc2c1c6c5c43                                  |
| Mol. weight [g/mol]: | 376.45                                                                            |
| CAS:                 | 190-87-4                                                                          |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | 991.38  | kJ/mol  | Joback Method  |
| hf            | 751.75  | kJ/mol  | Joback Method  |
| hfus          | 50.25   | kJ/mol  | Joback Method  |
| hvap          | 98.83   | kJ/mol  | Joback Method  |
| ie            | 7.27    | eV      | NIST Webbook   |
| ie            | 6.97    | eV      | NIST Webbook   |
| ie            | 6.97    | eV      | NIST Webbook   |
| ie            | 7.03    | eV      | NIST Webbook   |
| log10ws       | -12.93  |         | Crippen Method |
| logp          | 8.635   |         | Crippen Method |
| mcvol         | 282.180 | ml/mol  | McGowan Method |
| pc            | 1867.55 | kPa     | Joback Method  |
| tb            | 1059.82 | K       | Joback Method  |
| tc            | 1334.83 | K       | Joback Method  |
| tf            | 770.86  | K       | Joback Method  |
| vc            | 1.121   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 877.53 | J/mol×K | 1059.82         | Joback Method |
| cpg           | 900.56 | J/mol×K | 1105.65         | Joback Method |
| cpg           | 926.04 | J/mol×K | 1151.49         | Joback Method |
| cpg           | 954.48 | J/mol×K | 1197.32         | Joback Method |
| cpg           | 986.42 | J/mol×K | 1243.16         | Joback Method |

|       |           |         |         |               |
|-------|-----------|---------|---------|---------------|
| cpg   | 1022.37   | J/mol×K | 1288.99 | Joback Method |
| cpg   | 1062.84   | J/mol×K | 1334.83 | Joback Method |
| dvisc | 0.0277526 | Pa×s    | 770.86  | Joback Method |
| dvisc | 0.0275076 | Pa×s    | 819.02  | Joback Method |
| dvisc | 0.0272917 | Pa×s    | 867.18  | Joback Method |
| dvisc | 0.0270999 | Pa×s    | 915.34  | Joback Method |
| dvisc | 0.0269285 | Pa×s    | 963.50  | Joback Method |
| dvisc | 0.0267743 | Pa×s    | 1011.66 | Joback Method |
| dvisc | 0.0266349 | Pa×s    | 1059.82 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C190874&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C190874&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>ie:</b>      | Ionization energy                               |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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