

# Propane, 2,2-diallyloxy

<b>Other names:</b>	Propane, 2,2-di(2-propenyloxy) 3,3'-[(1-methylethylidene)bis(oxy)]bispropene
<b>Inchi:</b>	InChI=1S/C9H16O2/c1-5-7-10-9(3,4)11-8-6-2/h5-6H,1-2,7-8H2,3-4H3
<b>InchiKey:</b>	FMEUTDDJZQKSOU-UHFFFAOYSA-N
<b>Formula:</b>	C9H16O2
<b>SMILES:</b>	C=CCOC(C)(C)OCC=C
<b>Mol. weight [g/mol]:</b>	156.22
<b>CAS:</b>	35219-73-9

## Physical Properties

Property code	Value	Unit	Source
gf	-6.58	kJ/mol	Joback Method
hf	-251.42	kJ/mol	Joback Method
hfus	11.47	kJ/mol	Joback Method
hvap	37.81	kJ/mol	Joback Method
log10ws	-2.08		Crippen Method
logp	2.128		Crippen Method
mvol	140.810	ml/mol	McGowan Method
pc	2467.81	kPa	Joback Method
rinpol	893.00		NIST Webbook
rinpol	893.00		NIST Webbook
rinpol	903.00		NIST Webbook
tb	440.29	K	Joback Method
tc	620.69	K	Joback Method
tf	234.55	K	Joback Method
vc	0.526	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	294.01	J/mol×K	440.29	Joback Method
cpg	307.72	J/mol×K	470.36	Joback Method
cpg	320.83	J/mol×K	500.42	Joback Method
cpg	333.36	J/mol×K	530.49	Joback Method

cpg	345.32	J/molxK	560.56	Joback Method
cpg	356.73	J/molxK	590.63	Joback Method
cpg	367.61	J/molxK	620.69	Joback Method
dvisc	0.0038253	Paxs	234.55	Joback Method
dvisc	0.0016808	Paxs	268.84	Joback Method
dvisc	0.0008895	Paxs	303.13	Joback Method
dvisc	0.0005358	Paxs	337.42	Joback Method
dvisc	0.0003543	Paxs	371.71	Joback Method
dvisc	0.0002513	Paxs	406.00	Joback Method
dvisc	0.0001880	Paxs	440.29	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C35219739&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C35219739&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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