

# Hydrocoumarin

<b>Other names:</b>	1,2-Benzodihydropyrone 1,2-benzodihydropyrone (dihydrocoumarin) 2-Chromanone 2H-1-Benzopyran-2-one, 3,4-dihydro- 3,4-Dihydro-2H-1-benzopyran-2-one 3,4-Dihydrocoumarin Benzodihydropyrone Benzopyranone, dihydro- Chroman, 2-oxo- Coumarin, 3,4-dihydro- Dihydrobenzopyrone Hydrocinnamic acid, o-hydroxy-, «delta»-lactone Melilotin Melilotine Melilotol NCI-C55890 USAF DO-12 dihydrocoumarin
<b>Inchi:</b>	InChI=1S/C9H8O2/c10-9-6-5-7-3-1-2-4-8(7)11-9/h1-4H,5-6H2
<b>InchiKey:</b>	VMUXSMXIQBNMGZ-UHFFFAOYSA-N
<b>Formula:</b>	C9H8O2
<b>SMILES:</b>	O=C1CCc2ccccc2O1
<b>Mol. weight [g/mol]:</b>	148.16
<b>CAS:</b>	119-84-6

## Physical Properties

Property code	Value	Unit	Source
gf	-24.67	kJ/mol	Joback Method
hf	-186.75	kJ/mol	Joback Method
hfus	15.17	kJ/mol	Joback Method
hvap	69.90 ± 0.50	kJ/mol	NIST Webbook
log10ws	-2.06		Crippen Method
logp	1.538		Crippen Method
mcvol	110.490	ml/mol	McGowan Method
pc	4151.61	kPa	Joback Method
rinpol	1350.00		NIST Webbook
rinpol	1392.00		NIST Webbook

ripol	1359.00		NIST Webbook
ripol	1359.00		NIST Webbook
ripol	1359.00		NIST Webbook
ripol	1361.00		NIST Webbook
ripol	1398.00		NIST Webbook
ripol	1361.00		NIST Webbook
ripol	1359.00		NIST Webbook
ripol	1398.00		NIST Webbook
ripol	1350.00		NIST Webbook
ripol	2269.00		NIST Webbook
ripol	2286.00		NIST Webbook
ripol	2290.00		NIST Webbook
ripol	2286.00		NIST Webbook
ripol	2286.00		NIST Webbook
ripol	2286.00		NIST Webbook
ripol	2269.00		NIST Webbook
ripol	2314.00		NIST Webbook
tb	545.20	K	NIST Webbook
tc	799.04	K	Joback Method
tf	343.58	K	Joback Method
vc	0.409	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	251.63	J/mol×K	547.43	Joback Method
cpg	265.51	J/mol×K	589.36	Joback Method
cpg	278.43	J/mol×K	631.30	Joback Method
cpg	290.42	J/mol×K	673.23	Joback Method
cpg	301.51	J/mol×K	715.17	Joback Method
cpg	311.73	J/mol×K	757.10	Joback Method
cpg	321.13	J/mol×K	799.04	Joback Method
hvapt	69.90	kJ/mol	298.15	Experimental and computational thermochemistry of the isomers: Chromanone, 3-isochromanone, and dihydrocoumarin

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C119846&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C119846&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Experimental and computational thermochemistry of the isomers: chromanone, isochromanone, and dihydrocoumarin:</b>	<a href="https://www.doi.org/10.1016/j.jct.2008.08.012">https://www.doi.org/10.1016/j.jct.2008.08.012</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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