

Glutaric acid, monoamide, N-(2-octyl)-, decyl ester

Inchi:	InChI=1S/C23H45NO3/c1-4-6-8-10-11-12-13-15-20-27-23(26)19-16-18-22(25)24-21(3)17
InchiKey:	MXXFHBRBPXHATB-UHFFFAOYSA-N
Formula:	C23H45NO3
SMILES:	CCCCCCCCCOC(=O)CCCC(=O)NC(C)CCCCC
Mol. weight [g/mol]:	383.61

Physical Properties

Property code	Value	Unit	Source
gf	-133.11	kJ/mol	Joback Method
hf	-827.24	kJ/mol	Joback Method
hfus	61.29	kJ/mol	Joback Method
hvap	88.74	kJ/mol	Joback Method
log10ws	-7.39		Crippen Method
logp	6.316		Crippen Method
mcvol	353.920	ml/mol	McGowan Method
pc	920.50	kPa	Joback Method
rinpola	3157.00		NIST Webbook
tb	905.53	K	Joback Method
tc	1109.16	K	Joback Method
tf	508.72	K	Joback Method
vc	1.383	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1188.77	J/molxK	905.53	Joback Method
cpg	1208.29	J/molxK	939.47	Joback Method
cpg	1226.50	J/molxK	973.41	Joback Method
cpg	1243.45	J/molxK	1007.34	Joback Method
cpg	1259.18	J/molxK	1041.28	Joback Method
cpg	1273.75	J/molxK	1075.22	Joback Method
cpg	1287.19	J/molxK	1109.16	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U360858&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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