

# Ethanamine, 2,2-diethoxy-

<b>Other names:</b>	Acetaldehyde, amino-, diethyl acetal «alpha»-Aminoacetaldehyde diethyl acetal «beta», «beta»-Diethoxyethylamine Aminoacetaldehyde diethyl acetal Glycinaldehyde diethyl acetal 2-Aminoacetaldehyde diethyl acetal 2,2-Diethoxyethylamine 2,2-Diethoxyethanamine NSC 19501
<b>Inchi:</b>	InChI=1S/C6H15NO2/c1-3-8-6(5-7)9-4-2/h6H,3-5,7H2,1-2H3
<b>InchiKey:</b>	HJKLEAOXCZIMPI-UHFFFAOYSA-N
<b>Formula:</b>	C6H15NO2
<b>SMILES:</b>	CCOC(CN)OCC
<b>Mol. weight [g/mol]:</b>	133.19
<b>CAS:</b>	645-36-3

## Physical Properties

Property code	Value	Unit	Source
gf	-146.35	kJ/mol	Joback Method
hf	-403.10	kJ/mol	Joback Method
hfus	15.35	kJ/mol	Joback Method
hvap	44.02	kJ/mol	Joback Method
log10ws	-0.55		Crippen Method
logp	0.344		Crippen Method
mvol	117.120	ml/mol	McGowan Method
pc	3217.33	kPa	Joback Method
tb	436.00	K	NIST Webbook
tb	436.20	K	NIST Webbook
tc	636.84	K	Joback Method
tf	270.10	K	Joback Method
vc	0.430	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	257.16	J/mol×K	453.61	Joback Method
cpg	268.47	J/mol×K	484.15	Joback Method
cpg	279.43	J/mol×K	514.69	Joback Method
cpg	290.05	J/mol×K	545.22	Joback Method
cpg	300.31	J/mol×K	575.76	Joback Method
cpg	310.20	J/mol×K	606.30	Joback Method
cpg	319.73	J/mol×K	636.84	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C645363&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C645363&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/43-581-9/Ethanamine-2-2-diethoxy.pdf>

Generated by Cheméo on 2024-04-28 03:37:57.129619909 +0000 UTC m=+16564726.050197230.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.