

# 3-Fluoro-o-xylene

<b>Other names:</b>	1-Fluoro-2,3-dimethylbenzene 2,3-Dimethylfluorobenzene 3-Fluoro-1,2-xylene Benzene, 1-fluoro-2,3-dimethyl- o-Xylene, 3-fluoro-
<b>Inchi:</b>	InChI=1S/C8H9F/c1-6-4-3-5-8(9)7(6)2/h3-5H,1-2H3
<b>InchiKey:</b>	AWLDSXJCQWTJPC-UHFFFAOYSA-N
<b>Formula:</b>	C8H9F
<b>SMILES:</b>	Cc1cccc(F)c1C
<b>Mol. weight [g/mol]:</b>	124.16
<b>CAS:</b>	443-82-3

## Physical Properties

Property code	Value	Unit	Source
gf	-85.18	kJ/mol	Joback Method
hf	-190.97	kJ/mol	Joback Method
hfus	12.82	kJ/mol	Joback Method
hvap	36.19	kJ/mol	Joback Method
log10ws	-2.75		Crippen Method
logp	2.443		Crippen Method
mcvol	101.590	ml/mol	McGowan Method
pc	3341.24	kPa	Joback Method
tb	423.20	K	NIST Webbook
tc	619.18	K	Joback Method
tf	231.97	K	Joback Method
vc	0.394	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	186.01	J/molxK	418.35	Joback Method
cpg	197.13	J/molxK	451.82	Joback Method
cpg	207.71	J/molxK	485.29	Joback Method
cpg	217.79	J/molxK	518.76	Joback Method

cpg	227.36	J/mol×K	552.23	Joback Method
cpg	236.44	J/mol×K	585.71	Joback Method
cpg	245.06	J/mol×K	619.18	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.45466e+01
Coeff. B	-3.61799e+03
Coeff. C	-5.87860e+01
Temperature range (K), min.	312.52
Temperature range (K), max.	450.55

## Sources

**The Yaws Handbook of Vapor**

**Pressure:**

**Crippen Method:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C443823&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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