

# 5,9,13-Pentadecatrien-2-one, 6,10,14-trimethyl-

<b>Other names:</b>	5,9,13-Pentadecatriene-2-one, 6,10,14-trimethyl- Pentadeca-5,9,13-triene-2-one, 6,10,14-trimethyl- 6,10,14-Trimethyl-5,9,13-pentadecatrien-2-one 6,10,14-trimethylpentadeca-5,9,13-trien-2-one
<b>Inchi:</b>	InChI=1S/C18H30O/c1-15(2)9-6-10-16(3)11-7-12-17(4)13-8-14-18(5)19/h9,11,13H,6-8,1
<b>InchiKey:</b>	LTUMRKDLVGQMJU-UHFFFAOYSA-N
<b>Formula:</b>	C18H30O
<b>SMILES:</b>	<chem>CC(=O)CCC=C(C)CCC=C(C)CCC=C(C)C</chem>
<b>Mol. weight [g/mol]:</b>	262.43
<b>CAS:</b>	762-29-8

## Physical Properties

Property code	Value	Unit	Source
gf	186.77	kJ/mol	Joback Method
hf	-205.14	kJ/mol	Joback Method
hfus	40.65	kJ/mol	Joback Method
hvap	62.52	kJ/mol	Joback Method
log10ws	-6.20		Crippen Method
logp	5.775		Crippen Method
mcvol	253.150	ml/mol	McGowan Method
pc	1359.63	kPa	Joback Method
rinpol	1921.00		NIST Webbook
tb	677.23	K	Joback Method
tc	866.12	K	Joback Method
tf	285.43	K	Joback Method
vc	0.993	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	692.45	J/mol×K	677.23	Joback Method
cpg	710.76	J/mol×K	708.71	Joback Method
cpg	728.14	J/mol×K	740.19	Joback Method
cpg	744.65	J/mol×K	771.68	Joback Method

cpg	760.37	J/mol×K	803.16	Joback Method
cpg	775.34	J/mol×K	834.64	Joback Method
cpg	789.64	J/mol×K	866.12	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C762298&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C762298&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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