

# (Z)-9-Docosenoic acid, methyl ester

Other names:	9(Z)-Docosenoic acid, methyl ester
Inchi:	InChI=1S/C23H44O2/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-23(24)2
InchiKey:	NVCSYLJPLYJUNW-PFONDFGASA-N
Formula:	C23H44O2
SMILES:	CCCCCCCCCCCCC=CCCCCCCCC(=O)OC
Mol. weight [g/mol]:	352.59

## Physical Properties

Property code	Value	Unit	Source
gf	-10.92	kJ/mol	Joback Method
hf	-645.63	kJ/mol	Joback Method
hfus	58.31	kJ/mol	Joback Method
hvap	75.91	kJ/mol	Joback Method
log10ws	-8.17		Crippen Method
logp	7.757		Crippen Method
mcvol	338.070	ml/mol	McGowan Method
pc	902.89	kPa	Joback Method
rinpol	2479.00		NIST Webbook
tb	806.09	K	Joback Method
tc	988.24	K	Joback Method
tf	416.05	K	Joback Method
vc	1.327	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1068.83	J/molxK	806.09	Joback Method
cpg	1089.32	J/molxK	836.45	Joback Method
cpg	1108.77	J/molxK	866.81	Joback Method
cpg	1127.22	J/molxK	897.16	Joback Method
cpg	1144.72	J/molxK	927.52	Joback Method
cpg	1161.30	J/molxK	957.88	Joback Method
cpg	1177.00	J/molxK	988.24	Joback Method
dvisc	0.0011170	Paxs	416.05	Joback Method

dvisc	0.0004469	Paxs	481.06	Joback Method
dvisc	0.0002224	Paxs	546.06	Joback Method
dvisc	0.0001283	Paxs	611.07	Joback Method
dvisc	0.0000823	Paxs	676.08	Joback Method
dvisc	0.0000571	Paxs	741.08	Joback Method
dvisc	0.0000420	Paxs	806.09	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R98850&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R98850&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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