

# Carbonic acid, 2,2,2-trichloroethyl cyclohexyl ester

Inchi:	InChI=1S/C9H13Cl3O3/c10-9(11,12)6-14-8(13)15-7-4-2-1-3-5-7/h7H,1-6H2
InchiKey:	MDNHFAMHCYNWIL-UHFFFAOYSA-N
Formula:	C9H13Cl3O3
SMILES:	O=C(OCC(Cl)(Cl)Cl)OC1CCCCC1
Mol. weight [g/mol]:	275.56

## Physical Properties

Property code	Value	Unit	Source
gf	-322.52	kJ/mol	Joback Method
hf	-607.76	kJ/mol	Joback Method
hfus	20.05	kJ/mol	Joback Method
hvap	59.48	kJ/mol	Joback Method
log10ws	-4.09		Crippen Method
logp	3.842		Crippen Method
mcvol	176.840	ml/mol	McGowan Method
pc	2657.03	kPa	Joback Method
rinqol	1624.00		NIST Webbook
tb	632.64	K	Joback Method
tc	864.53	K	Joback Method
tf	385.14	K	Joback Method
vc	0.650	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	422.98	J/molxK	632.64	Joback Method
cpg	437.63	J/molxK	671.29	Joback Method
cpg	451.19	J/molxK	709.94	Joback Method
cpg	463.68	J/molxK	748.58	Joback Method
cpg	475.14	J/molxK	787.23	Joback Method
cpg	485.60	J/molxK	825.88	Joback Method
cpg	495.07	J/molxK	864.53	Joback Method
dvisc	0.0019954	Paxs	385.14	Joback Method
dvisc	0.0010577	Paxs	426.39	Joback Method

dvisc	0.0006271	Paxs	467.64	Joback Method
dvisc	0.0004047	Paxs	508.89	Joback Method
dvisc	0.0002789	Paxs	550.14	Joback Method
dvisc	0.0002024	Paxs	591.39	Joback Method
dvisc	0.0001532	Paxs	632.64	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357895&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357895&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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