

# Diethylmalonic acid, 3-bromobenzyl octadecyl ester

Inchi:	InChI=1S/C32H53BrO4/c1-4-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-25-36-30(34)3
InchiKey:	RNGARFMARSIFOE-UHFFFAOYSA-N
Formula:	C32H53BrO4
SMILES:	CCCCCCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCc1cccc(Br)c1
Mol. weight [g/mol]:	581.66

## Physical Properties

Property code	Value	Unit	Source
gf	-129.34	kJ/mol	Joback Method
hf	-950.77	kJ/mol	Joback Method
hfus	75.73	kJ/mol	Joback Method
hvap	113.21	kJ/mol	Joback Method
log10ws	-11.46		Crippen Method
logp	10.103		Crippen Method
mcvol	470.360	ml/mol	McGowan Method
pc	694.71	kPa	Joback Method
rmpol	3483.00		NIST Webbook
tb	1178.73	K	Joback Method
tc	1473.01	K	Joback Method
tf	695.88	K	Joback Method
vc	1.819	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1664.17	J/molxK	1178.73	Joback Method
cpg	1745.00	J/molxK	1423.97	Joback Method
cpg	1731.52	J/molxK	1374.92	Joback Method
cpg	1716.94	J/molxK	1325.87	Joback Method
cpg	1701.02	J/molxK	1276.82	Joback Method
cpg	1683.51	J/molxK	1227.78	Joback Method
cpg	1757.63	J/molxK	1473.01	Joback Method
dvisc	0.0000046	Paxs	1178.73	Joback Method
dvisc	0.0000061	Paxs	1098.26	Joback Method

dvisc	0.0000085	Paxs	1017.78	Joback Method
dvisc	0.0000125	Paxs	937.31	Joback Method
dvisc	0.0000198	Paxs	856.83	Joback Method
dvisc	0.0000344	Paxs	776.36	Joback Method
dvisc	0.0000680	Paxs	695.88	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U368425&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U368425&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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