

# Benzeneacetaldehyde, 4-methyl-

<b>Other names:</b>	p-Tolyl-acetaldehyde
<b>Inchi:</b>	InChI=1S/C9H10O/c1-8-2-4-9(5-3-8)6-7-10/h2-5,7H,6H2,1H3
<b>InchiKey:</b>	CIXAYNMKFFQEFU-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O
<b>SMILES:</b>	<chem>Cc1ccc(CC=O)cc1</chem>
<b>Mol. weight [g/mol]:</b>	134.18
<b>CAS:</b>	104-09-6

## Physical Properties

Property code	Value	Unit	Source
gf	28.16	kJ/mol	Joback Method
hf	-89.61	kJ/mol	Joback Method
hfus	15.01	kJ/mol	Joback Method
hvap	45.29	kJ/mol	Joback Method
log10ws	-2.03		Crippen Method
logp	1.736		Crippen Method
mcvol	115.480	ml/mol	McGowan Method
pc	3493.01	kPa	Joback Method
rinpol	1119.90		NIST Webbook
rinpol	1119.90		NIST Webbook
ripol	1772.00		NIST Webbook
ripol	1772.00		NIST Webbook
tb	494.70	K	NIST Webbook
tc	699.60	K	Joback Method
tf	272.13	K	Joback Method
vc	0.449	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	235.04	J/mol×K	485.64	Joback Method
cpg	289.07	J/mol×K	663.94	Joback Method
cpg	279.56	J/mol×K	628.28	Joback Method
cpg	269.43	J/mol×K	592.62	Joback Method

cpg	258.65	J/molxK	556.96	Joback Method
cpg	247.20	J/molxK	521.30	Joback Method
cpg	297.98	J/molxK	699.60	Joback Method
dvisc	0.0002820	Paxs	485.64	Joback Method
dvisc	0.0003503	Paxs	450.06	Joback Method
dvisc	0.0004517	Paxs	414.47	Joback Method
dvisc	0.0006110	Paxs	378.88	Joback Method
dvisc	0.0008797	Paxs	343.30	Joback Method
dvisc	0.0013782	Paxs	307.71	Joback Method
dvisc	0.0024281	Paxs	272.13	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	369.20	K	1.30	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C104096&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C104096&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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