

# Cholan-24-oic acid, 3,6,7-tris(acetyloxy)-, methyl ester,

Other names: Methyl 3,6,7-tris(acetyloxy)cholan-24-oate-  
(3«alpha»,5«beta»,6«alpha»,7«alpha»)-  
Hyocholic acid, tris-acetateoxy, methyl ester

Hyocholic acid, acetate-methyl ester

<b>Inchi:</b>	InChI=1S/C31H48O8/c1-17(8-11-26(35)36-7)22-9-10-23-27-24(13-15-30(22,23)5)31(6)1
<b>InchiKey:</b>	SCZJGLWPRVUGAT-UHFFFAOYSA-N
<b>Formula:</b>	C31H48O8
<b>SMILES:</b>	COC(=O)CCC(C)C1CCC2C3C(OC(C)=O)C(OC(C)=O)C4CC(OC(C)=O)CCC4(C)C3CCC
<b>Mol. weight [g/mol]:</b>	548.71
<b>CAS:</b>	2616-70-8

## Physical Properties

Property code	Value	Unit	Source
gf	-602.72	kJ/mol	Joback Method
hf	-1498.81	kJ/mol	Joback Method
hfus	59.54	kJ/mol	Joback Method
hvap	117.19	kJ/mol	Joback Method
log10ws	-6.23		Crippen Method
logp	5.249		Crippen Method
mcvol	433.970	ml/mol	McGowan Method
pc	832.90	kPa	Joback Method
rinsol	3452.00		NIST Webbook
tb	1234.17	K	Joback Method
tc	1518.97	K	Joback Method
tf	789.29	K	Joback Method
vc	1.639	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1827.98	J/molxK	1234.17	Joback Method
cpg	1870.45	J/molxK	1281.64	Joback Method
cpg	1914.97	J/molxK	1329.10	Joback Method
cpg	1962.05	J/molxK	1376.57	Joback Method
cpg	2012.18	J/molxK	1424.04	Joback Method

cpg	2065.86	J/mol×K	1471.51	Joback Method
cpg	2123.59	J/mol×K	1518.97	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2616708&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2616708&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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