

# N,N-Dimethylpivalamide

<b>Other names:</b>	N,N,2,2-Tetramethylpropionamide N,N,2,2-Tetramethylpropanamide 2,2-Dimethylpropionic acid dimethylamide N,N-Dimethyl-tert-butylcarboxamide
<b>Inchi:</b>	InChI=1S/C7H15NO/c1-7(2,3)6(9)8(4)5/h1-5H3
<b>InchiKey:</b>	RLVGHTRVYWUWSH-UHFFFAOYSA-N
<b>Formula:</b>	C7H15NO
<b>SMILES:</b>	CN(C)C(=O)C(C)(C)C
<b>Mol. weight [g/mol]:</b>	129.20
<b>CAS:</b>	24331-71-3

## Physical Properties

Property code	Value	Unit	Source
affp	927.10	kJ/mol	NIST Webbook
basg	895.20	kJ/mol	NIST Webbook
basg	899.00 ± 5.00	kJ/mol	NIST Webbook
chs	-4557.10 ± 1.90	kJ/mol	NIST Webbook
gf	-7.24	kJ/mol	Joback Method
hf	-286.10 ± 2.10	kJ/mol	NIST Webbook
hfs	-341.20 ± 2.10	kJ/mol	NIST Webbook
hfus	11.09	kJ/mol	Joback Method
hsub	55.10 ± 0.40	kJ/mol	NIST Webbook
hvap	38.67	kJ/mol	Joback Method
log10ws	-0.86		Crippen Method
logp	1.121		Crippen Method
mcvol	121.040	ml/mol	McGowan Method
pc	3052.41	kPa	Joback Method
rinpol	1023.00		NIST Webbook
rinpol	1023.00		NIST Webbook
tb	422.64	K	Joback Method
tc	608.97	K	Joback Method
tf	253.47	K	Joback Method
vc	0.441	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	246.91	J/mol×K	422.64	Joback Method
cpg	260.70	J/mol×K	453.69	Joback Method
cpg	273.75	J/mol×K	484.75	Joback Method
cpg	286.08	J/mol×K	515.80	Joback Method
cpg	297.72	J/mol×K	546.86	Joback Method
cpg	308.71	J/mol×K	577.91	Joback Method
cpg	319.07	J/mol×K	608.97	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	348.50 ± 0.50	K	2.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C24331713&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C24331713&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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