

Benzenamine, 4-(phenylazo)-

Other names:	4-(Phenylazo)benzenamine
	4-Amino-1,1'-azobenzene
	4-Aminoazobenzene
	4-Aminoazobenzol
	4-Benzeneazoaniline
	4-Phenylazoaniline
	Aminoazobenzene
	Aniline Yellow
	Aniline, p-(phenylazo)-
	Azobenzene, 4-amino-
	Benzenamine, 4-(2-phenyldiazenyl)-
	Brasilazina Oil Yellow G
	C.I. 11000
	C.I. Solvent Blue 7
	C.I. Solvent Yellow 1
	Cellitazol R
	Ceres Yellow R
	Fast Spirit Yellow
	Fast Spirit Yellow AAB
	Fat Yellow AAB
	Induline R
	Oil Soluble Aniline Yellow
	Oil Yellow 2G
	Oil Yellow AAB
	Oil Yellow AB
	Oil Yellow AN
	Oil Yellow B
	Oil Yellow R
	Oil-Sol. Aniline Yellow
	Organol Yellow
	Organol Yellow 2A
	P-(PHENYLAZO)ANILINE
	P-AMINOAZOBENZENE
	P-AMINOAZOBENZOL
	P-AMINODIPHENYLIMIDE
	Paraphenolazo aniline
	Solvent Yellow 1
	Somalia Yellow 2G
	Stearix Brown 4R
	Sudan Yellow R

Sudan Yellow RA
USAF EK-1375
Zlut anilinovala
Zlut rozpoustedlova 1
p-(Phenolazo)aniline
p-(Phenylazo)phenylamine
p-Amimoazobenzene
p-Aminodiphenyldiimide

Inchi: InChI=1S/C12H11N3/c13-10-6-8-12(9-7-10)15-14-11-4-2-1-3-5-11/h1-9H,13H2
InchiKey: QPQKUYVVSJWQSDY-UHFFFAOYSA-N
Formula: C12H11N3
SMILES: Nc1ccc(N=Nc2ccccc2)cc1
Mol. weight [g/mol]: 197.24
CAS: 60-09-3

Physical Properties

Property code	Value	Unit	Source
chs	-6617.40	kJ/mol	NIST Webbook
hf	251.59	kJ/mol	Joback Method
hfs	323.00	kJ/mol	NIST Webbook
hvap	64.83	kJ/mol	Joback Method
log10ws	-3.15		Crippen Method
logp	3.684		Crippen Method
mcvol	158.060	ml/mol	McGowan Method
pc	2690.21	kPa	Joback Method
tb	633.00	K	NIST Webbook
tc	1029.51	K	Joback Method
tf	398.00 ± 1.00	K	NIST Webbook
tf	399.00	K	NIST Webbook

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cps	276.10	J/mol×K	323.00	NIST Webbook
hfust	21.70	kJ/mol	398.20	NIST Webbook
hfust	21.70	kJ/mol	398.20	NIST Webbook
hsubt	110.90 ± 1.70	kJ/mol	364.50	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52070e+01
Coeff. B	-5.57270e+03
Coeff. C	-1.06713e+02
Temperature range (K), min.	480.23
Temperature range (K), max.	669.86

Sources

The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
KDB:	https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1495
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C60093&Units=SI

Legend

chs:	Standard solid enthalpy of combustion
cps:	Solid phase heat capacity
hf:	Enthalpy of formation at standard conditions
hfs:	Solid phase enthalpy of formation at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hsubt:	Enthalpy of sublimation at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure

pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point

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