

# 2-Propoxy-tetrahydropyran

|                      |                                                        |
|----------------------|--------------------------------------------------------|
| Inchi:               | InChI=1S/C8H16O2/c1-2-6-9-8-5-3-4-7-10-8/h8H,2-7H2,1H3 |
| InchiKey:            | FSQXTURTITWAAE-UHFFFAOYSA-N                            |
| Formula:             | C8H16O2                                                |
| SMILES:              | CCCOC1CCCCO1                                           |
| Mol. weight [g/mol]: | 144.21                                                 |
| CAS:                 | 6581-64-2                                              |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -150.19 | kJ/mol  | Joback Method  |
| hf            | -418.35 | kJ/mol  | Joback Method  |
| hfus          | 17.48   | kJ/mol  | Joback Method  |
| hvap          | 40.75   | kJ/mol  | Joback Method  |
| log10ws       | -1.85   |         | Crippen Method |
| logp          | 1.940   |         | Crippen Method |
| mcvol         | 124.460 | ml/mol  | McGowan Method |
| pc            | 3062.55 | kPa     | Joback Method  |
| rinpol        | 1005.00 |         | NIST Webbook   |
| rinpol        | 1005.00 |         | NIST Webbook   |
| tb            | 451.36  | K       | Joback Method  |
| tc            | 651.27  | K       | Joback Method  |
| tf            | 236.10  | K       | Joback Method  |
| vc            | 0.456   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 270.72 | J/mol×K | 451.36          | Joback Method |
| cpg           | 287.23 | J/mol×K | 484.68          | Joback Method |
| cpg           | 303.02 | J/mol×K | 518.00          | Joback Method |
| cpg           | 318.10 | J/mol×K | 551.32          | Joback Method |
| cpg           | 332.47 | J/mol×K | 584.64          | Joback Method |
| cpg           | 346.15 | J/mol×K | 617.95          | Joback Method |
| cpg           | 359.13 | J/mol×K | 651.27          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0059882 | Pa×s | 236.10 | Joback Method |
| dvisc | 0.0025373 | Pa×s | 271.98 | Joback Method |
| dvisc | 0.0013133 | Pa×s | 307.85 | Joback Method |
| dvisc | 0.0007799 | Pa×s | 343.73 | Joback Method |
| dvisc | 0.0005111 | Pa×s | 379.61 | Joback Method |
| dvisc | 0.0003603 | Pa×s | 415.48 | Joback Method |
| dvisc | 0.0002685 | Pa×s | 451.36 | Joback Method |

## Sources

|                        |                                                                                                                                             |
|------------------------|---------------------------------------------------------------------------------------------------------------------------------------------|
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                       |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6581642&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6581642&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                   |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                           |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                       |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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