

# Diethylmalonic acid, decyl 4-methoxyphenyl ester

Inchi:	InChI=1S/C24H38O5/c1-5-8-9-10-11-12-13-14-19-28-22(25)24(6-2,7-3)23(26)29-21-17-18
InchiKey:	YTXWTWKCFZPUSK-UHFFFAOYSA-N
Formula:	C24H38O5
SMILES:	CCCCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccc(OC)cc1
Mol. weight [g/mol]:	406.56

## Physical Properties

Property code	Value	Unit	Source
gf	-316.02	kJ/mol	Joback Method
hf	-944.20	kJ/mol	Joback Method
hfus	50.92	kJ/mol	Joback Method
hvap	91.38	kJ/mol	Joback Method
log10ws	-6.80		Crippen Method
logp	6.091		Crippen Method
mcvol	346.010	ml/mol	McGowan Method
pc	1032.57	kPa	Joback Method
rinsol	2738.00		NIST Webbook
tb	951.95	K	Joback Method
tc	1166.26	K	Joback Method
tf	568.15	K	Joback Method
vc	1.327	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1149.27	J/molxK	951.95	Joback Method
cpg	1217.25	J/molxK	1130.54	Joback Method
cpg	1206.33	J/molxK	1094.82	Joback Method
cpg	1194.12	J/molxK	1059.10	Joback Method
cpg	1180.57	J/molxK	1023.39	Joback Method
cpg	1165.63	J/molxK	987.67	Joback Method
cpg	1226.92	J/molxK	1166.26	Joback Method
dvisc	0.0000174	Paxs	951.95	Joback Method
dvisc	0.0000230	Paxs	887.98	Joback Method

dvisc	0.0000316	Paxs	824.02	Joback Method
dvisc	0.0000459	Paxs	760.05	Joback Method
dvisc	0.0000713	Paxs	696.08	Joback Method
dvisc	0.0001213	Paxs	632.12	Joback Method
dvisc	0.0002324	Paxs	568.15	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369845&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369845&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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