

# Isophthalic acid, di(2,2,3,3,4,4,5,5-octafluoropentyl) ester

<b>Inchi:</b>	InChI=1S/C18H10F16O4/c19-11(20)15(27,28)17(31,32)13(23,24)5-37-9(35)7-2-1-3-8(4-
<b>InchiKey:</b>	NYOUDFDPEYLGEO-UHFFFAOYSA-N
<b>Formula:</b>	C18H10F16O4
<b>SMILES:</b>	O=C(OCC(F)(F)C(F)(F)C(F)(F)C(F)F)c1cccc(C(=O)OCC(F)(F)C(F)(F)C(F)(F)C(F)F)c1
<b>Mol. weight [g/mol]:</b>	594.24

## Physical Properties

Property code	Value	Unit	Source
gf	-3369.18	kJ/mol	Joback Method
hf	-3880.21	kJ/mol	Joback Method
hfus	39.35	kJ/mol	Joback Method
hvap	55.29	kJ/mol	Joback Method
log10ws	-7.82		Crippen Method
logp	6.342		Crippen Method
mvol	283.920	ml/mol	McGowan Method
pc	1022.04	kPa	Joback Method
rinpol	1828.00		NIST Webbook
rinpol	1828.00		NIST Webbook
tb	763.54	K	Joback Method
tc	935.40	K	Joback Method
tf	469.84	K	Joback Method
vc	1.194	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	895.73	J/molxK	763.54	Joback Method
cpg	906.65	J/molxK	792.18	Joback Method
cpg	916.69	J/molxK	820.83	Joback Method
cpg	925.91	J/molxK	849.47	Joback Method
cpg	934.41	J/molxK	878.11	Joback Method
cpg	942.27	J/molxK	906.76	Joback Method
cpg	949.58	J/molxK	935.40	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U356603&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U356603&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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