

cis-Phenanthrene, 9,10-dihydro-9-methyl-9,10-diol, 3-methoxy

Inchi:	InChI=1S/C16H16O3/c1-16(18)14-6-4-3-5-11(14)13-9-10(19-2)7-8-12(13)15(16)17/h3-9,
InchiKey:	ISRUVUJGAGZJCM-CVEARBPZSA-N
Formula:	C16H16O3
SMILES:	COc1ccc2c(c1)-c1cccc1C(C)(O)C2O
Mol. weight [g/mol]:	256.30

Physical Properties

Property code	Value	Unit	Source
gf	-39.22	kJ/mol	Joback Method
hf	-297.74	kJ/mol	Joback Method
hfus	28.48	kJ/mol	Joback Method
hvap	91.80	kJ/mol	Joback Method
log10ws	-4.43		Crippen Method
logp	2.617		Crippen Method
mvol	195.530	ml/mol	McGowan Method
pc	2912.39	kPa	Joback Method
rmpol	2350.00		NIST Webbook
tb	838.60	K	Joback Method
tc	1053.42	K	Joback Method
tf	545.47	K	Joback Method
vc	0.734	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	588.06	J/mol×K	838.60	Joback Method
cpg	600.97	J/mol×K	874.40	Joback Method
cpg	613.78	J/mol×K	910.21	Joback Method
cpg	626.63	J/mol×K	946.01	Joback Method
cpg	639.66	J/mol×K	981.82	Joback Method
cpg	653.03	J/mol×K	1017.62	Joback Method
cpg	666.86	J/mol×K	1053.42	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R109514&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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