

# (cis-4,5-Methylenehexyl)-cyclopropane

<b>Inchi:</b>	InChI=1S/C10H18/c1-8-7-10(8)4-2-3-9-5-6-9/h8-10H,2-7H2,1H3/t8-,10+/m0/s1
<b>InchiKey:</b>	GBESKWVEEIKQFO-WCBMZHEXSA-N
<b>Formula:</b>	C10H18
<b>SMILES:</b>	CC1CC1CCCC1CC1
<b>Mol. weight [g/mol]:</b>	138.25

## Physical Properties

Property code	Value	Unit	Source
gf	147.11	kJ/mol	Joback Method
hf	-124.47	kJ/mol	Joback Method
hfus	19.00	kJ/mol	Joback Method
hvap	37.37	kJ/mol	Joback Method
log10ws	-3.07		Crippen Method
logp	3.223		Crippen Method
mvol	130.040	ml/mol	McGowan Method
pc	2627.15	kPa	Joback Method
rinpol	1018.80		NIST Webbook
rinpol	1015.30		NIST Webbook
tb	437.01	K	Joback Method
tc	626.35	K	Joback Method
tf	234.10	K	Joback Method
vc	0.508	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	286.74	J/mol×K	437.01	Joback Method
cpg	305.06	J/mol×K	468.57	Joback Method
cpg	322.35	J/mol×K	500.12	Joback Method
cpg	338.68	J/mol×K	531.68	Joback Method
cpg	354.08	J/mol×K	563.23	Joback Method
cpg	368.62	J/mol×K	594.79	Joback Method
cpg	382.35	J/mol×K	626.35	Joback Method
dvisc	0.0006447	Paxs	234.10	Joback Method

dvisc	0.0006543	Paxs	267.92	Joback Method
dvisc	0.0006618	Paxs	301.74	Joback Method
dvisc	0.0006679	Paxs	335.56	Joback Method
dvisc	0.0006729	Paxs	369.37	Joback Method
dvisc	0.0006771	Paxs	403.19	Joback Method
dvisc	0.0006807	Paxs	437.01	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R137856&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R137856&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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