

# Formic acid, 4-methylpent-2-yl ester

<b>Inchi:</b>	InChI=1S/C7H14O2/c1-6(2)4-7(3)9-5-8/h5-7H,4H2,1-3H3
<b>InchiKey:</b>	VJKXDWCVZONWBR-UHFFFAOYSA-N
<b>Formula:</b>	C7H14O2
<b>SMILES:</b>	CC(C)CC(C)OC=O
<b>Mol. weight [g/mol]:</b>	130.18

## Physical Properties

Property code	Value	Unit	Source
gf	-201.34	kJ/mol	Joback Method
hf	-416.17	kJ/mol	Joback Method
hfus	10.32	kJ/mol	Joback Method
hvap	39.53	kJ/mol	Joback Method
log10ws	-1.49		Crippen Method
logp	1.594		Crippen Method
mcvol	116.930	ml/mol	McGowan Method
pc	3028.94	kPa	Joback Method
rinpola	828.00		NIST Webbook
tb	429.76	K	Joback Method
tc	609.26	K	Joback Method
tf	202.88	K	Joback Method
vc	0.451	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	240.27	J/molxK	429.76	Joback Method
cpg	293.40	J/molxK	579.34	Joback Method
cpg	283.58	J/molxK	549.42	Joback Method
cpg	273.35	J/molxK	519.51	Joback Method
cpg	262.72	J/molxK	489.59	Joback Method
cpg	251.70	J/molxK	459.68	Joback Method
cpg	302.84	J/molxK	609.26	Joback Method
dvisc	0.0002613	Paxs	429.76	Joback Method
dvisc	0.0003547	Paxs	391.95	Joback Method

dvisc	0.0005137	Paxs	354.13	Joback Method
dvisc	0.0008130	Paxs	316.32	Joback Method
dvisc	0.0014574	Paxs	278.51	Joback Method
dvisc	0.0031385	Paxs	240.69	Joback Method
dvisc	0.0089962	Paxs	202.88	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U367940&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U367940&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/44-139-9/Formic-acid-4-methylpent-2-yl-ester.pdf>

Generated by Cheméo on 2024-04-28 07:34:58.613244043 +0000 UTC m=+16578947.533821354.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.