

# Benzenemethanol, 4-bromo-«alpha»-methyl-

<b>Other names:</b>	Benzyl alcohol, p-bromo-«alpha»-methyl- 4-Bromo-«alpha»-methylbenzenemethanol p-Bromo-«alpha»-methylbenzyl alcohol 4-Bromophenylmethylcarbinol p-Bromophenyl methyl carbinol 1-(4-Bromophenyl)ethanol 4-Bromo-«alpha»-methylbenzylalcohol 4-bromo-alpha-methylbenzyl alcohol
<b>Inchi:</b>	InChI=1S/C8H9BrO/c1-6(10)7-2-4-8(9)5-3-7/h2-6,10H,1H3
<b>InchiKey:</b>	XTDTYSBVMQBIBT-UHFFFAOYSA-N
<b>Formula:</b>	C8H9BrO
<b>SMILES:</b>	CC(O)c1ccc(Br)cc1
<b>Mol. weight [g/mol]:</b>	201.06
<b>CAS:</b>	5391-88-8

## Physical Properties

Property code	Value	Unit	Source
gf	-5.68	kJ/mol	Joback Method
hf	-114.57	kJ/mol	Joback Method
hfus	15.98	kJ/mol	Joback Method
hvap	59.07	kJ/mol	Joback Method
log10ws	-3.16		Crippen Method
logp	2.502		Crippen Method
mcvol	123.190	ml/mol	McGowan Method
pc	4438.52	kPa	Joback Method
rinpol	1314.70		NIST Webbook
tb	572.00	K	Joback Method
tc	789.10	K	Joback Method
tf	324.48	K	Joback Method
vc	0.451	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	256.95	J/molxK	572.00	Joback Method
cpg	266.72	J/molxK	608.18	Joback Method
cpg	275.85	J/molxK	644.37	Joback Method
cpg	284.37	J/molxK	680.55	Joback Method
cpg	292.31	J/molxK	716.73	Joback Method
cpg	299.71	J/molxK	752.92	Joback Method
cpg	306.62	J/molxK	789.10	Joback Method
dvisc	0.0067893	Paxs	324.48	Joback Method
dvisc	0.0023207	Paxs	365.73	Joback Method
dvisc	0.0009861	Paxs	406.99	Joback Method
dvisc	0.0004905	Paxs	448.24	Joback Method
dvisc	0.0002745	Paxs	489.49	Joback Method
dvisc	0.0001681	Paxs	530.75	Joback Method
dvisc	0.0001105	Paxs	572.00	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	393.20	K	0.90	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5391888&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5391888&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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