

# (Z)-3-hexenyl pentafluorobenzoate

<b>Inchi:</b>	InChI=1S/C13H11F5O2/c1-2-3-4-5-6-20-13(19)7-8(14)10(16)12(18)11(17)9(7)15/h3-4H,1
<b>InchiKey:</b>	XRWDLUDVMKQGG-ARJAWSKDSA-N
<b>Formula:</b>	C13H11F5O2
<b>SMILES:</b>	CCC=CCCOC(=O)c1c(F)c(F)c(F)c(F)c1F
<b>Mol. weight [g/mol]:</b>	294.22

## Physical Properties

Property code	Value	Unit	Source
gf	-1004.91	kJ/mol	Joback Method
hf	-1240.60	kJ/mol	Joback Method
hfus	39.91	kJ/mol	Joback Method
hvap	55.15	kJ/mol	Joback Method
log10ws	-5.32		Crippen Method
logp	3.895		Crippen Method
mcvol	182.260	ml/mol	McGowan Method
pc	1830.98	kPa	Joback Method
ripol	1404.00		NIST Webbook
ripol	1413.00		NIST Webbook
ripol	1408.00		NIST Webbook
ripol	1408.00		NIST Webbook
ripol	1414.00		NIST Webbook
ripol	1743.00		NIST Webbook
ripol	1747.00		NIST Webbook
ripol	1746.00		NIST Webbook
ripol	1753.00		NIST Webbook
ripol	1753.00		NIST Webbook
tb	625.22	K	Joback Method
tc	800.76	K	Joback Method
tf	395.32	K	Joback Method
vc	0.750	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	460.65	J/mol×K	625.22	Joback Method
cpg	471.98	J/mol×K	654.48	Joback Method
cpg	482.79	J/mol×K	683.73	Joback Method
cpg	493.09	J/mol×K	712.99	Joback Method
cpg	502.88	J/mol×K	742.25	Joback Method
cpg	512.17	J/mol×K	771.50	Joback Method
cpg	520.97	J/mol×K	800.76	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R311910&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R311910&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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