

# Diethylmalonic acid, dodecyl 2-methoxyethyl ester

Inchi:	InChI=1S/C22H42O5/c1-5-8-9-10-11-12-13-14-15-16-17-26-20(23)22(6-2,7-3)21(24)27-1
InchiKey:	HUYDZSCMLWGLOW-UHFFFAOYSA-N
Formula:	C22H42O5
SMILES:	CCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCCOC
Mol. weight [g/mol]:	386.57

## Physical Properties

Property code	Value	Unit	Source
gf	-435.64	kJ/mol	Joback Method
hf	-1127.98	kJ/mol	Joback Method
hfus	52.08	kJ/mol	Joback Method
hvap	83.99	kJ/mol	Joback Method
log10ws	-5.60		Crippen Method
logp	5.447		Crippen Method
mcvol	341.590	ml/mol	McGowan Method
pc	954.96	kPa	Joback Method
rinsol	2411.00		NIST Webbook
tb	874.53	K	Joback Method
tc	1070.79	K	Joback Method
tf	506.67	K	Joback Method
vc	1.323	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1124.45	J/molxK	874.53	Joback Method
cpg	1143.14	J/molxK	907.24	Joback Method
cpg	1160.56	J/molxK	939.95	Joback Method
cpg	1176.74	J/molxK	972.66	Joback Method
cpg	1191.71	J/molxK	1005.37	Joback Method
cpg	1205.50	J/molxK	1038.08	Joback Method
cpg	1218.13	J/molxK	1070.79	Joback Method
dvisc	0.0004155	Paxs	506.67	Joback Method
dvisc	0.0001979	Paxs	567.98	Joback Method

dvisc	0.0001089	Paxs	629.29	Joback Method
dvisc	0.0000667	Paxs	690.60	Joback Method
dvisc	0.0000442	Paxs	751.91	Joback Method
dvisc	0.0000312	Paxs	813.22	Joback Method
dvisc	0.0000231	Paxs	874.53	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370679&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370679&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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