

# Phenol, 2,6-bis(1,1-dimethylethyl)-4-ethoxy-

<b>Inchi:</b>	InChI=1S/C16H26O2/c1-8-18-11-9-12(15(2,3)4)14(17)13(10-11)16(5,6)7/h9-10,17H,8H2
<b>InchiKey:</b>	YQQQXXUABFURLN-UHFFFAOYSA-N
<b>Formula:</b>	C16H26O2
<b>SMILES:</b>	CCOc1cc(C(C)(C)C)c(O)c(C(C)(C)C)c1
<b>Mol. weight [g/mol]:</b>	250.38
<b>CAS:</b>	5442-35-3

## Physical Properties

Property code	Value	Unit	Source
gf	-76.95	kJ/mol	Joback Method
hf	-487.01	kJ/mol	Joback Method
hfus	22.60	kJ/mol	Joback Method
hvap	67.64	kJ/mol	Joback Method
log10ws	-4.19		Crippen Method
logp	4.386		Crippen Method
mcvol	224.280	ml/mol	McGowan Method
pc	1918.62	kPa	Joback Method
rinpol	1575.00		NIST Webbook
tb	698.70	K	Joback Method
tc	918.63	K	Joback Method
tf	460.33	K	Joback Method
vc	0.785	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	643.65	J/molxK	698.70	Joback Method
cpg	722.49	J/molxK	881.97	Joback Method
cpg	708.43	J/molxK	845.32	Joback Method
cpg	693.63	J/molxK	808.66	Joback Method
cpg	677.97	J/molxK	772.01	Joback Method
cpg	661.35	J/molxK	735.35	Joback Method
cpg	735.92	J/molxK	918.63	Joback Method
dvisc	0.0000076	Paxs	698.70	Joback Method

dvisc	0.0000115	Paxs	658.97	Joback Method
dvisc	0.0000185	Paxs	619.24	Joback Method
dvisc	0.0000317	Paxs	579.52	Joback Method
dvisc	0.0000587	Paxs	539.79	Joback Method
dvisc	0.0001202	Paxs	500.06	Joback Method
dvisc	0.0002782	Paxs	460.33	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5442353&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5442353&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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