

# Limonen-10-ol

<b>Other names:</b>	p-mentha-1,8-dien-9-ol p-mentha-1,8-dien-10-ol 1,8-menthadien-9-ol 1,8-p-Menthadien-9-ol Limonene-10-ol p-menth-1,8-dien-9-ol p-Mentha-1,8-diene-10-ol p-Menta-1,8-dien-10-ol p-Menthadien-1,8-dien-9-ol 4-methyl-«beta»-methylenecyclohex-3-ene-1-ethanol
<b>Inchi:</b>	InChI=1S/C10H16O/c1-8-3-5-10(6-4-8)9(2)7-11/h3,10-11H,2,4-7H2,1H3
<b>InchiKey:</b>	UIMAEYMKYMNCGW-UHFFFAOYSA-N
<b>Formula:</b>	C10H16O
<b>SMILES:</b>	<chem>C=C(CO)C1CC=C(C)CC1</chem>
<b>Mol. weight [g/mol]:</b>	152.23
<b>CAS:</b>	3269-90-7

## Physical Properties

Property code	Value	Unit	Source
gf	20.57	kJ/mol	Joback Method
hf	-185.69	kJ/mol	Joback Method
hfus	15.82	kJ/mol	Joback Method
hvap	55.33	kJ/mol	Joback Method
log10ws	-2.63		Crippen Method
logp	2.281		Crippen Method
mcvol	138.170	ml/mol	McGowan Method
pc	3042.32	kPa	Joback Method
ripol	1239.00		NIST Webbook
ripol	1289.00		NIST Webbook
ripol	1274.00		NIST Webbook
ripol	1987.00		NIST Webbook
ripol	1979.00		NIST Webbook
ripol	1979.00		NIST Webbook
ripol	1939.00		NIST Webbook
ripol	1980.00		NIST Webbook
ripol	1962.00		NIST Webbook
ripol	1996.00		NIST Webbook

ripol	2008.00		NIST Webbook
tb	540.63	K	Joback Method
tc	736.52	K	Joback Method
tf	268.22	K	Joback Method
vc	0.515	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	330.83	J/mol×K	540.63	Joback Method
cpg	345.48	J/mol×K	573.28	Joback Method
cpg	359.35	J/mol×K	605.93	Joback Method
cpg	372.48	J/mol×K	638.57	Joback Method
cpg	384.90	J/mol×K	671.22	Joback Method
cpg	396.63	J/mol×K	703.87	Joback Method
cpg	407.69	J/mol×K	736.52	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3269907&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3269907&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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