

# Diethylmalonic acid, ethyl 2-isopropylphenyl ester

|                      |                                                                                   |
|----------------------|-----------------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C18H26O4/c1-6-18(7-2,16(19)21-8-3)17(20)22-15-12-10-9-11-14(15)13(4)5/h9 |
| InchiKey:            | ISUOTBCDLNJIKO-UHFFFAOYSA-N                                                       |
| Formula:             | C18H26O4                                                                          |
| SMILES:              | CCOC(=O)C(CC)(CC)C(=O)Oc1ccccc1C(C)C                                              |
| Mol. weight [g/mol]: | 306.40                                                                            |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -263.98 | kJ/mol  | Joback Method  |
| hf            | -693.42 | kJ/mol  | Joback Method  |
| hfus          | 30.66   | kJ/mol  | Joback Method  |
| hvap          | 75.23   | kJ/mol  | Joback Method  |
| log10ws       | -4.52   |         | Crippen Method |
| logp          | 4.085   |         | Crippen Method |
| mcvol         | 255.600 | ml/mol  | McGowan Method |
| pc            | 1586.01 | kPa     | Joback Method  |
| rinpol        | 1861.00 |         | NIST Webbook   |
| tb            | 791.81  | K       | Joback Method  |
| tc            | 1001.25 | K       | Joback Method  |
| tf            | 463.30  | K       | Joback Method  |
| vc            | 0.967   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 763.36    | J/mol×K | 791.81          | Joback Method |
| cpg           | 779.50    | J/mol×K | 826.72          | Joback Method |
| cpg           | 794.50    | J/mol×K | 861.62          | Joback Method |
| cpg           | 808.40    | J/mol×K | 896.53          | Joback Method |
| cpg           | 821.22    | J/mol×K | 931.44          | Joback Method |
| cpg           | 833.02    | J/mol×K | 966.34          | Joback Method |
| cpg           | 843.81    | J/mol×K | 1001.25         | Joback Method |
| dvisc         | 0.0007844 | Pa×s    | 463.30          | Joback Method |
| dvisc         | 0.0003954 | Pa×s    | 518.05          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0002272 | Pa×s | 572.80 | Joback Method |
| dvisc | 0.0001438 | Pa×s | 627.55 | Joback Method |
| dvisc | 0.0000980 | Pa×s | 682.31 | Joback Method |
| dvisc | 0.0000707 | Pa×s | 737.06 | Joback Method |
| dvisc | 0.0000533 | Pa×s | 791.81 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369965&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369965&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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