

# N1N1-dimethyl-N2-ortho-methoxy-phenylformamide

Inchi:	InChI=1S/C10H14N2O/c1-12(2)8-11-9-6-4-5-7-10(9)13-3/h4-8H,1-3H3/b11-8+
InchiKey:	IAXYUZNWRIJSIY-DHZHZOJOSA-N
Formula:	C10H14N2O
SMILES:	COc1ccccc1N=CN(C)C
Mol. weight [g/mol]:	178.23

## Physical Properties

Property code	Value	Unit	Source
hf	-7.14	kJ/mol	Joback Method
hvap	48.56	kJ/mol	Joback Method
log10ws	-1.64		Crippen Method
logp	1.917		Crippen Method
mcvol	149.530	ml/mol	McGowan Method
pc	2535.37	kPa	Joback Method
rinpol	1557.00		NIST Webbook
tb	571.40	K	Joback Method
tc	790.60	K	Joback Method

## Sources

NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R164006&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R164006&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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