

# Cyclopentene, octachloro-

<b>Other names:</b>	Octachlorocyclopentene Perchlorocyclopentene NSC 4736
<b>Inchi:</b>	InChI=1S/C5Cl8/c6-1-2(7)4(10,11)5(12,13)3(1,8)9
<b>InchiKey:</b>	DMZRCHJVVAKCAX-UHFFFAOYSA-N
<b>Formula:</b>	C5Cl8
<b>SMILES:</b>	C1C=C(Cl)C(Cl)(Cl)C(Cl)(Cl)C1(Cl)Cl
<b>Mol. weight [g/mol]:</b>	343.68
<b>CAS:</b>	706-78-5

## Physical Properties

Property code	Value	Unit	Source
chl	-1979.60 ± 1.80	kJ/mol	NIST Webbook
gf	-88.86	kJ/mol	Joback Method
hf	-96.70 ± 3.30	kJ/mol	NIST Webbook
hfl	-176.80 ± 2.00	kJ/mol	NIST Webbook
hfus	19.91	kJ/mol	Joback Method
hvap	80.00 ± 1.00	kJ/mol	NIST Webbook
log10ws	-5.70		Crippen Method
logp	5.211		Crippen Method
mcvol	164.070	ml/mol	McGowan Method
pc	3306.75	kPa	Joback Method
rinpol	1618.80		NIST Webbook
rinpol	1601.80		NIST Webbook
tb	556.20	K	NIST Webbook
tc	912.76	K	Joback Method
tf	485.39	K	Joback Method
vc	0.626	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	258.35	J/mol×K	629.02	Joback Method
cpg	261.98	J/mol×K	676.31	Joback Method

cpg	265.84	J/mol×K	723.60	Joback Method
cpg	270.54	J/mol×K	770.89	Joback Method
cpg	276.68	J/mol×K	818.18	Joback Method
cpg	284.87	J/mol×K	865.47	Joback Method
cpg	295.70	J/mol×K	912.76	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C706785&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C706785&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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