

# 4-(CH2Cl)-C6H4-CN

|                      |                                                                            |
|----------------------|----------------------------------------------------------------------------|
| Other names:         | 3-Cyanobenzylchloride<br>4-(CH2Cl)-C6H4-CN<br>4-(chloromethyl)benzonitrile |
| Inchi:               | InChI=1S/C8H6ClN/c9-5-7-1-3-8(6-10)4-2-7/h1-4H,5H2                         |
| InchiKey:            | LOQLDQJTSMKBJU-UHFFFAOYSA-N                                                |
| Formula:             | C8H6ClN                                                                    |
| SMILES:              | N#Cc1ccc(CCl)cc1                                                           |
| Mol. weight [g/mol]: | 151.60                                                                     |

## Physical Properties

| Property code | Value   | Unit   | Source             |
|---------------|---------|--------|--------------------|
| af            | 0.4412  |        | Cheméo Relay (1.0) |
| affp          | 812.80  | kJ/mol | NIST Webbook       |
| basg          | 782.10  | kJ/mol | NIST Webbook       |
| gf            | 240.51  | kJ/mol | Joback Method      |
| hf            | 129.94  | kJ/mol | Cheméo Relay (1.0) |
| hfus          | 15.83   | kJ/mol | Joback Method      |
| hvap          | 67.88   | kJ/mol | Cheméo Relay (1.0) |
| ie            | 9.78    | eV     | Cheméo Relay (1.0) |
| log10ws       | -2.72   |        | Cheméo Relay (1.0) |
| logp          | 2.297   |        | Crippen Method     |
| mcvol         | 113.440 | ml/mol | McGowan Method     |
| pc            | 3333.53 | kPa    | Joback Method      |
| tb            | 527.16  | K      | Cheméo Relay (1.0) |
| tc            | 769.28  | K      | Cheméo Relay (1.0) |
| tf            | 340.71  | K      | Cheméo Relay (1.0) |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 222.76 | J/mol×K | 553.61          | Joback Method |
| cpg           | 232.04 | J/mol×K | 593.20          | Joback Method |
| cpg           | 240.67 | J/mol×K | 632.78          | Joback Method |
| cpg           | 248.68 | J/mol×K | 672.37          | Joback Method |

|     |        |         |        |               |
|-----|--------|---------|--------|---------------|
| cpg | 256.09 | J/mol×K | 711.95 | Joback Method |
| cpg | 262.95 | J/mol×K | 751.54 | Joback Method |
| cpg | 269.27 | J/mol×K | 791.13 | Joback Method |

## Sources

|                                  |                                                                                                                                           |
|----------------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Crippen Method:</b>           | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Cheméo Relay (1.0):</b>       |                                                                                                                                           |
| <b>Cheméo Relay (1.0, beta):</b> |                                                                                                                                           |
| <b>NIST Webbook:</b>             | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C874862&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C874862&amp;Units=SI</a> |
| <b>Joback Method:</b>            | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b>           | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |
| <b>Crippen Method:</b>           | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>af:</b>      | Acentric Factor                                 |
| <b>affp:</b>    | Proton affinity                                 |
| <b>basg:</b>    | Gas basicity                                    |
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>ie:</b>      | Ionization energy                               |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |

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